8.72 (d, $4, J=7 \mathrm{~Hz}, \mathrm{H}-4^{\prime}$ ), 8.82 (s, 8, $\beta$-pyrrolic H ), 9.09 (s, 4, H-2'). The porphyrin $m-\mathrm{NO}_{2}(\mathrm{TPP}) \mathrm{H}_{2}(11 \mathrm{~b})(2.72 \mathrm{~g}, 3.42 \mathrm{mmol}), 137 \mathrm{~mL}$ of concentrated aqueous HCl , and stannous chloride dihydrate $(12.3 \mathrm{~g}$, 54.8 mmol ) were combined as previously reported. ${ }^{29}$ The resulting mixture was made basic ( 140 mL of $\mathrm{NH}_{4} \mathrm{OH}$ ), and 50 mL of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was added. After the suspension was thoroughly mixed, the resulting precipitate was collected by filtration. The dark solid collected was crushed to a powder and extracted with THF ( $4 \times 150 \mathrm{~mL}$ ). The resulting THF extract was filtered through a pad of silica gel. The filtrate was reduced to $30 \mathrm{~mL}, 50 \mathrm{~mL}$ of $\mathrm{CHCl}_{3}$ was added, and the solution was again reduced to 30 mL when another 50 mL of $\mathrm{CHCl}_{3}$ was added. After the solution was reduced to a final 20 mL , the precipitate was filtered and washed with $\mathrm{CHCl}_{3}$ to give $2.00 \mathrm{~g}(87 \%)$ of $m-\mathrm{NH}_{2}$. (TPP) $\mathrm{H}_{2}$ (11c) as a fine purple microcrystals: $R_{f} 0.46$ (3:1 THF/hexanes); ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ) $\delta-2.96$ (s, 2, pyrrolic NH), 5.48 (br s, 8, $\mathrm{NH}_{2}$ ), $7.01\left(\mathrm{~d}, 4, J=7 \mathrm{~Hz}, \mathrm{H}-4^{\prime}\right), 7.36-7.46\left(\mathrm{~m}, 12, \mathrm{H}-2^{\prime}, \mathrm{H}-5^{\prime}\right.$, and $\mathrm{H}-6^{\prime}$ ), 8.92 (s, 8, $\beta$-pyrrolic H); FABMS $m / 2675$ (caled for $\mathrm{C}_{44} \mathrm{H}_{34} \mathrm{~N}_{8}$ ( $\mathrm{M}^{+}$) 675).

5,10,15,20-Tetrakis[ $\boldsymbol{m}$-( $p$-tolylsulfonamido) phenyl]porphyrin (13) ( $m$-(NHTs)(TPP) $\mathrm{H}_{2}$ ). A mixture of $m-\mathrm{NH}_{2}$ (TPP) $\mathrm{H}_{2}$ (11a) ( 100 mg , 0.148 mmol ), $p$-toluenesulfonyl chloride ( $565 \mathrm{mg}, 2.96 \mathrm{mmol}$ ), and triethylamine ( $0.52 \mathrm{~mL}, 3.7 \mathrm{mmol}$ ) in 30 mL of THF was stirred at room temperature for 72 h , then 10 mL of methanol was added, and the solution was stirred for an additional 14 h . The mixture was then diluted with ethyl acetate followed by the usual workup. The residue was dissolved in a minimal amount of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and this solution was layered with benzene. The resulting red-purple crystals were collected by filtration and dried at $110^{\circ} \mathrm{C}(0.2 \mathrm{mmHg})$ over $\mathrm{P}_{2} \mathrm{O}_{5}$ for 14 h to give 166 mg ( $87 \%$ ) of $m$-(NHTs)(TPP) $\mathrm{H}_{2}$ (13): $R_{f} 0.62$ (3:1 THF/hexanes); ${ }^{1} \mathrm{H}$ NMR (DMSO- $d_{6}$ ) $\delta-3.17\left(\mathrm{~s}, 2\right.$, pyrrolic NH), $2.34\left(\mathrm{~s}, 12\right.$, tosyl $\left.\mathrm{CH}_{3}\right)$, $7.46(\mathrm{~d}, 8, J=7.5 \mathrm{~Hz}$, tosyl $\mathrm{m}-\mathrm{H}), 7.65(\mathrm{br} \mathrm{s}, 4), 7.70-7.82(\mathrm{~m}, 16)$, 7.87-7.96 (m, 4), 8.42-8.51 (m, 8, $\beta$-pyrrolic H ), 10.55-10.62 (m, 4, tosyl NH); FABMS $m / z 1290$ (calcd for $\mathrm{C}_{72} \mathrm{H}_{58} \mathrm{~N}_{8} \mathrm{O}_{8} \mathrm{~S}_{4}\left(\mathrm{M}^{+}\right) 1290$ ).

Tetrakis[ $m, m^{\prime}$-[methylene-( $p$-tolylsulfonyl)imino]]-strati-bis-(5,10,15,20-tetraphenylporphyrin) (14b) ( $\left.m, m \cdot{ }^{\prime} \cdot \mathrm{TsNCH}_{2}-(\mathrm{TPP}) \mathrm{H}_{2}\right)_{2}$ ). A mixture of $m$-(NHTs)(TPP) (13) $(160 \mathrm{mg}, 0.124 \mathrm{mmol}), m-\mathrm{CH}_{2} \mathrm{Br}$ (TPP) $\mathrm{H}_{2}$ (2) ( 122 mg .0 .124 mmol ), and $\mathrm{Cs}_{2} \mathrm{CO}_{3}(242 \mathrm{mg}, 0.744 \mathrm{mmol})$ in 125 mL of DMF was stirred at room temperalure for 14 h , then diluted with 50 mL of $\mathrm{CHCl}_{3}$, and worked up in the usual manner. The residue obtained was dried further at $110^{\circ} \mathrm{C}(0.05 \mathrm{mmHg})$ over $\mathrm{P}_{2} \mathrm{O}_{5}$
for 45 min . The residue ( 220 mg ) was subjected to flash chromatography on silica, and elution with $100: 1 \mathrm{CHCl}_{3} /$ methanol provided 55 mg of a purple residue.

To simplify the preparative TLC, ${ }^{24}$ the material thus obtained was metalated in the following manner. ${ }^{37}$ To a refluxing $\mathrm{CHCl}_{3}$ solution (8 mL ) of this crude residue was added 5 mL of a methanolic solution containing 15 mg of potassium acetate ( 0.15 mmol ) and 15 mg of Zn -$(\mathrm{OAc})_{2}-2 \mathrm{H}_{2} \mathrm{O}(0.68 \mathrm{mmol})$. This mixture was refluxed for 30 min , and then the solvent was evaporated. The residue was dissolved in $\mathrm{CHCl}_{3}$ and filtered over Celite. The filtrate was subjected to preparative TLC on a $0.5 \times 200 \times 200 \mathrm{~mm}$ silica plate eluting with $100: 1 \mathrm{CHCl}_{3} /$ methanol, and the second most nonpolar magenta band was isolated. This residue was dissolved in 0.3 mL of trifluoroacetic acid, and this solution was stirred for 15 min at room temperature, then diluted with $\mathrm{CHCl}_{3}$, and washed with $5 \%$ aqueous $\mathrm{NH}_{4} \mathrm{OH}$ solution followed by the standard workup to give 1.3 mg ( $1 \%$ ) of $m, m^{\prime}-\mathrm{Ts} \mathrm{NCH}_{2}-\left((\mathrm{TPP}) \mathrm{H}_{2}\right)_{2}$ (14) as a purple solid: $R_{f} 0.46\left(100: 1 \mathrm{CHCl}_{3} /\right.$ methanol); ${ }^{1} \mathrm{H}$ NMR $\delta-4.11,-4.07$ ( $\mathrm{s}, 2$ each, pyrrolic NH), $2.25\left(\mathrm{~s}, 12\right.$, tosyl $\left.\mathrm{CH}_{3}\right), 5.19\left(\mathrm{~s}, 8, \mathrm{CH}_{2}\right), 6.95$ $\left(\mathrm{s}, 4, \mathrm{H}-2^{\prime \prime \prime}\right), 7.26(\mathrm{~d}, 8, J=8 \mathrm{~Hz}$, tosyl $m-\mathrm{H}), 7.50(\mathrm{~d}, 4, J=8 \mathrm{~Hz}$, $\left.\mathrm{H}-4^{\prime \prime \prime}\right), 7.60^{\prime}\left(\mathrm{t}, 4, J=8 \mathrm{~Hz}, \mathrm{H}-5^{\prime \prime \prime}\right), 7.66\left(\mathrm{~d}, 4, J=8 \mathrm{~Hz}, \mathrm{H}-4^{\prime \prime}\right), 7.76$ $\left(\mathrm{t}, 4, J=8 \mathrm{~Hz}, \mathrm{H}-5^{\prime \prime}\right), 7.78\left(\mathrm{~s}, 4, \mathrm{H}-2^{\prime \prime}\right), 7.79(\mathrm{~d}, 8, J=8 \mathrm{~Hz}$, tosyl o-H), $7.96\left(\mathrm{~d}, 4, J=8 \mathrm{~Hz}, \mathrm{H}-6^{\prime \prime \prime}\right), 8.25$ (br s, $8, \beta^{\prime}$-pyrrolic H), 8.35 (d, $4, J$ $=8 \mathrm{~Hz}, \mathrm{H}-6^{\prime \prime}$ ), 8.43 (s, 8, $\beta$-pyrrolic H); IR $\nu 3320(\mathrm{w}, \mathrm{N}-\mathrm{H}), 1600$ $(\mathrm{m}, \mathrm{C}=\mathrm{C}) 1270,1170$, and $1100\left(\mathrm{~s}, \mathrm{SO}_{2}\right) \mathrm{cm}^{-1}$; UV/vis $\lambda_{\max }\left(\epsilon \times 10^{-3}\right.$ $\mathrm{cm}^{-1} \mathrm{M}^{-1}$ ) $415 \mathrm{~nm}(314), 517$ (16.6), 552 (7.71), 593 (5.69), 650 (4.03); FABMS $m / z 1953$ (calcd for $\mathrm{C}_{120} \mathrm{H}_{88} \mathrm{~N}_{12} \mathrm{O}_{8} \mathrm{~S}_{4}\left(\mathrm{M}^{+}\right)$1953).

Acknowledgment. We express our gratitude to Protos Corp. of Emeryville, CA, for a grant in support of this study.

Supplementary Material Available: ${ }^{1} \mathrm{H}$ NMR spectra of compounds 2, 3b, and $15\left(\mathrm{X}=\mathrm{CF}_{3} \mathrm{CO}_{2}^{-}\right.$, the aromatic region as compared to 3c) each in $\mathrm{CDCl}_{3}$ and 14 in $\mathrm{CDCl}_{3}$ and $\mathrm{CD}_{2} \mathrm{Cl}_{2}$ (5 pages). Ordering information is given on any current masthead page.
(37) Porphyrins and Metalloporphyrins; Sminh, K. M., Ed.; Elsevier Scientific Publishing Co.: New York, 1975; p 798.

# Highly Felkin-Anh Selective Hiyama Additions of Chiral Allylic Bromides to Aldehydes. Application to the First Synthesis of Nephromopsinic Acid and Its Enantiomer 

Johann Mulzer, ${ }^{*,}{ }^{\dagger}$ Lars Kattner, ${ }^{\dagger}$ Achim R. Strecker, ${ }^{\dagger}$ Christian Schröder, ${ }^{\dagger}{ }^{\dagger}{ }^{\$}$ Jürgen Buschmann, ${ }^{\ddagger}$ Christian Lehmann, ${ }^{\ddagger}$ and Peter Luger ${ }^{\ddagger, \perp}$<br>Contribution from the Institut für Organische Chemie der Freien Universität Berlin, Takustrasse 3, D-1000 Berlin 33, FRG, and Institut für Kristallographie der Freien Universität Berlin, Takustrasse 6, D-1000 Berlin 33. FRG. Received October 15, 1990


#### Abstract

The chromium(II)-mediated addition ("Hiyama reaction") of the chiral allylic bromides $13,15,19,22,24$, and 27 to achiral and chiral aldehydes proceeds with high Felkin-Anh selectivity with respect to the stereocenter at $\mathrm{C}_{\gamma}$ in the bromide (Table II). By double stereodifferentiation experiments (Tables III/IV) it was shown that the bromide is the stereodominating component in the addition. The methodology was applied to the first synthesis of nephromopsinic acid ( - )-69, found in the lichen species nephromopsis stracheyi, and its enantiomer.


Allyl transfer reactions from reagents $\mathbf{1 a}-\mathrm{g}$ to aldehydes have acquired a central importance in natural product synthesis, due to their high regio- and stereochemical predictability. ${ }^{1}$ The $\mathrm{C}-\mathrm{C}$

[^0]connection regioselectively occurs at the $\gamma$-position of the double bond with concomitant allylic shift and migration of X to the aldehyde oxygen, from where it is removed by hydrolysis. The simple diastereoselection (syn or anti configuration at the newly
(1) (a) Hoffmann, R. W. Angew. Chem. Int. Ed. Engl. 1982, 21, 555. (b) Yamamolo, Y. Acc. Chem. Res. 1987, 20. 243. (c) Hoppe, D. Angew. Chem., Int. Ed. Engl. 1984. 23. 932.



created stereocenters) can be controlled by an appropriate choice of $X$ and the $E / Z$ geometry of 1 (Table I).
With respect to enantioselectivity, high inductions have been achieved for $1 \mathbf{a} / \mathbf{b}^{\mathbf{2}}$ and $\mathbf{1 c} / \mathrm{d}^{9}$ by chiral ligands at the boron atom, without changing the simple diastereoselection of the addition. A different approach has been developed by Hoffmann, ${ }^{10}$ who employed $\alpha$-chiral allylboronic esters 6. The addition proceeds with efficient chirality transfer to the $\gamma$-position to furnish stereoisomers 7 and 8 in ratios of better than 90:10. The chiral substitution in the diol part of 6 has no effect on the enantioselectivity of the reaction. High enantiocontrol has been reported also for titanium reagents $1 e^{11 a-c}$ with appropriate chiral ligands R , for chiral allylic stannanes, ${ }^{\text {,1d }}$ and crotylmolybdenum complexes. ${ }^{1 \mathrm{le}}$


We studied the Hiyama reaction of the $\delta$-chiral allylic bromides 9 with aldehydes. Presuming that the reaction continues to favor the anti diastereomers we had to expect 10 and 11 as the main products. The new questions to resolve then were as follows. (a) Which influence do chiral centers in 9 have on the ratio of 10:11;
(2) (a) Brown, H. C.; Bhal, K. S.; Randad, R. S. J. Org. Chem. 1989, 54, 1570. (b) Brown, H. C.; Bhat, K. S. J. Am. Chem. Soc. 1986, 108, 5919. (c) Corey, E. J.; Yu. C.-M.: Kim, S. S. J. Am. Chem. Soc. 1989, 111, 5495.
(3) Yamaguchi, M.; Mukaiyama, T. Chem. Lett. 1980, 993.
(4) (a) Hoffmann, R. W.: Zeiss, H.-J. J. Org. Chem. 1981, 46, 1309. (b) Andersen, M. W.; Hildebrandt, B.: Köster, G.; Hoffmann, R. W. Chem. Ber. 1989, 122, 1777.
(5) (a) Sato, F.; Iida, K.: Iijima, S.; Moriya, H.; Sato, M. J. Chem. Soc., Chem. Commun. 1981, 1140. (b) Widler, L.; Seebach, D. Helv. Chim. Acta 1982, 65, 1085. (c) Reetz, M. T.; Steinbach, R.; Weslermann, R. P.; Wenderolh, B. Chem. Ber. 1985, 1/8, 1441.
(6) Yamamoto, Y.; Yalagai, H.; Ishihara, Y.; Maeda, N.; Maruyama, K. Tetrahedron 1984, $40,2239$.
(7) (a) Okude, Y.; Hirano, S.; Hiyama, T., Nozaki, H. J. Am. Chem. Soc. 1977, 99, 3179. (b) Hiyama, T.; Kimura, K.; Nozaki, H. Tetrahedron Lett. 1981, 22, 1037. (c) Hiyama, T.; Okude, Y.; Kimura, K.; Nozaki, H. Bull. Chem. Soc. Jpn. 1982, 55, 561.
(8) Buse, C. T.; Healhcock. C. H. Tetrahedron Lett. 1978. 1685.
(9) (a) Roush, W. R.; Halterman, R. L. J. Am. Chem. Soc. 1986. 108, 294. (b) Roush, W. R.; Walis, A. E.; Hoong, L. K. J. Am. Chem. Soc. 1985, 107, 8186. (c) Roush, W. R.; Hoong, L. K.; Palmer, M. A. J.; Straub. J. A.: Palkowitz, A. D. J. Org. Chem. 1990, 55, 4117.
(10) Hoffmann, R. W.; Ditrich, K.; Köster, G.; Stürmer, R. Chem. Ber. 1989. 122, 1783 and literalure cited therein.
(11) (a) Riediker, M.; Duthaler, R. O. Angew. Chem. Int. Ed. Engl. 1989, 28, 494. (b) Hoppe, D.; Zschage, O. Angew. Chem. Int. Ed. Engl. 1989, 28, 69. (c) Roder, H.; Helmchen, G.; Pelers, E. M.; Peters, K.; von Schnering, H.-G. Angew. Chem. Int. Ed. Engl. 1984, 23, 898. (d) Jephcole, V. J.; Pratt, A. J.; Thomas, E. J. J. Chem. Soc., Chem. Commun. 1984, 800 . (e) Faller, J. W.; John, J. A.: Mazzieri, M. R. Tetrahedron Lett. 1989, $30,1769$.

Table I. Selectivities for Allyl Transfer Reagents 1

|  |  | config |  |  |
| :--- | :--- | :--- | :---: | :--- |
| $\mathbf{1}$ | X | diastereoselection | ref |  |
| $\mathbf{a}$ | $\mathrm{BR}_{2}$ | $Z$ | syn | 2 |
| $\mathbf{b}$ | $\mathrm{BR}_{2}$ | $E$ | anti | 2,3 |
| $\mathbf{c}$ | $\mathrm{~B}(\mathrm{OR})_{2}$ | $Z$ | syn | 4 |
| $\mathbf{d}$ | $\mathrm{~B}(\mathrm{OR})_{2}$ | $E$ | anti | 4 |
| $\mathbf{e}$ | $\mathrm{Ti}(\mathrm{OR})_{x} \mathrm{R}_{y}$ | unknown | anti | 5 |
| $\mathbf{f}$ | $\mathrm{SnR}{ }_{3}$ | $E$ or $Z$ | syn | 6 |
| $\mathbf{g}$ | $\mathrm{Cr}^{11} \mathrm{R}_{n}^{\prime}$ | unknown | anti | 7,8 |



Figure 1. Crystal structure of 44.
especially, can the effect of the $\delta$-center be modified by the introduction of additional stereocenters in the $\epsilon$ - and $\zeta$-positions? (b) Which one of the established models developed for the interpretation of 1,2 -induction in acyclic systems ${ }^{12}$ may be used to interpret the stereochemical outcome? (c) How can the reaction be applied in natural product synthesis?


Substituents: $L$ ( large), $M$ (medium) , $S$ (small)
Synthesis of the Model Allylic Bromides (Scheme I). As shown in Scheme I, the following three types of Hiyama reagents were prepared: (a) those with one stereogenic center in the $\delta$-position (13 and 15), (b) those with two stereogenic centers in the $\delta$ - and $\epsilon$-position, both in syn-19 and anti-22 configurations, and (c) those with three stereogenic centers in the $\delta-, \epsilon-$, and $\zeta$-positions, in anti,anti-24 and anti-syn-27 configurations. In all cases, the same standard sequence was employed, starting from the aldehydes (12a, 14a, 18a, 21a, 23a, and 26a) which were converted into the ( $E$ )-acrylic esters ( $\mathbf{1 2 b}, \mathbf{1 4 b}, \mathbf{1 8 b}, \mathbf{2 1 b}, \mathbf{2 3 b}$, and 26b), reduced to the allylic alcohols (12c, 14c, 18c, 21c, 23c, 26c), and then brominated with $\mathrm{PBr}_{3}$. The aldehydes were partly known (12a, ${ }^{13}$ 14a ${ }^{14}$ ) or were prepared from known precursors ( $16,{ }^{15} 20,{ }^{16} 25{ }^{16}$ ) as described in Scheme I.

## Stereochemical Results

All Hiyama additions were performed in anhydrous THF at 0 to $-5^{\circ} \mathrm{C}$ over a period of $1-3$ days in chemical yields between 55 and $90 \%$. The chromium(II) chloride was prepared in situ from chromium(III) chloride and lithium aluminum hydride or purchased from Aldrich. The results were the same in both cases.

[^1]Scheme Ia




$\left.\begin{array}{ll}21 & a \\ & b \\ & c\end{array}\right] b$
22


$12,14,18,21,23,26$

a (a) (EtO) ${ }_{2} \mathrm{POCH}_{2} \mathrm{CO}_{2} \mathrm{Et} / \mathrm{NaH}, \mathrm{THF}, 0-22{ }^{\circ} \mathrm{C}, 24 \mathrm{~h}, 70-95 \%$; (b) DIBAH. toluene/ $\mathrm{Et}_{2} \mathrm{O},-40$ to $-5^{\circ} \mathrm{C}, 4 \mathrm{~h}, 78-91 \%$; (c) $\mathrm{PBr}_{3}, \mathrm{Et}_{2} \mathrm{O}$, -25 to $22^{\circ} \mathrm{C}, 2 \mathrm{~h}, 79-98 \%$; (d) Red-Al, toluene, $0-22^{\circ} \mathrm{C}, 24 \mathrm{~h}, 85 \%$ : (e) $\mathrm{BnCl} / \mathrm{NaH}, \mathrm{DMF}, 0-22{ }^{\circ} \mathrm{C}, 24 \mathrm{~h}, 94-95 \%$; (f) $\mathrm{O}_{3}$ and then $\mathrm{PPh}_{3}$, $\mathrm{CH}_{2} \mathrm{Cl}_{2},-78$ to $22^{\circ} \mathrm{C}, 84-85 \%$; (g) p - $\mathrm{TsOH}, \mathrm{MeOH}, 22^{\circ} \mathrm{C}, 24 \mathrm{~h}$, $82 \%$; (h) $\mathrm{H}_{5} \mathrm{lO}_{6}$, THF, $22{ }^{\circ} \mathrm{C}, 20 \mathrm{~min}$; and then 2 equiv of $\mathrm{LiAlH}_{4}$, $0-22^{\circ} \mathrm{C}, 24 \mathrm{~h}, 87 \%$.

Achiral Aldehydes (Table II). Benzaldehyde and tetradecanal were chosen as achiral aldehydes. Bromides 13 and 15 reacted with benzaldehyde under moderate diastereofacial selectivity to furnish, almost independent of the O -protective group, diaste-

${ }^{a}$ (a) $\mathrm{Na} / \mathrm{NH}_{3}, \mathrm{THF},-40^{\circ} \mathrm{C}, 30 \mathrm{~min}, 74-91 \%$; (b) $\mathrm{H}_{5} \mathrm{IO}_{6}, \mathrm{THF}, 22$ ${ }^{\circ} \mathrm{C}, 20 \mathrm{~min}$ and then $\mathrm{LiAlH}_{4}, \mathrm{THF}, 0-22^{\circ} \mathrm{C}, 24 \mathrm{~h}, 95 \%$; (c) $p$ - TsOH $\mathrm{MeOH}, 22^{\circ} \mathrm{C}, 24 \mathrm{~h}, 94 \%$.
reomers $28 / 29$ and $\mathbf{3 0} / 31$, respectively, in ratios of $83: 17$ and 82:18, according to HPLC analysis. The bromides with two stereocenters (19/22) showed a significantly higher selectivity ( $>90: 10$ ) and furnished $32 / 33$ and $34 / 35$ as the only products. The presence of a third stereocenter (24/27) led to a further increase in selection; diastereomers $36 / 37,38 / 39$, and $40 / 41$ were formed in a ratio of 96:4. On the $\Delta \Delta G^{*}$ scale each stereocenter accounts for $0.4-0.5 \mathrm{kcal} / \mathrm{mol}$.

The configuration at the newly created stereocenters of the major diastereomers were determined as shown in Scheme II. 28 and 30 were deprotected to afford the same diol 42 , which was converted into the acetonide 43 and analyzed by NOE difference spectroscopy. In particular, the strong interaction between $3 \cdot \mathrm{CH}_{3}$ and $\mathrm{H} \cdot \mathrm{I}$ and 2 ( $11-14 \%$ NOE each) clearly indicated the configuration shown. 32 was debenzylated to give the crystalline triol 44, which was submitted to a single-crystal X-ray analysis (Figure 1). Remarkably, an intramolecular hydrogen bond forming a seven-membered ring ( $\mathrm{HO}-\mathrm{H}$ distance $1.71 \AA$ ) can be recognized. 34 and 36 were degraded via 45 and 46 , respectively, to furnish diol 42 , identical in all respects ( ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, HPLC) with the material obtained from 28 and $\mathbf{3 0}$. The structures of $\mathbf{3 8}$ and 40, eventually, followed from the conversion of these adducts into $[(-)$ - and ( + )]-nephromopsinic acid (69) (vide infra). The configurations of the minor diastereomers were not rigorously established but assigned on the basis of the well-established simple diastereoselection of the Hiyama reaction. ${ }^{8}$
In conclusion, all major diastereomers have an all-syn arrangement of the $\beta^{\prime}-\mathrm{OH}, \gamma$-vinyl, and $\delta$-methyl substituents. This means that (1) it is only the $\delta$-center which determines the configuration at the newly created stereocenters ( $\gamma$ and $\beta^{\prime}$ ), (2) the simple diastereoselection of the Hiyama reaction is not affected by the presence of chiral centers in the allylic bromide, and (3) additional stereocenters in the $\epsilon$ - and $\zeta$-positions of the bromide increase the diastereofacial selectivity but have no influence on the sense of the asymmetric induction.
Stereochemical Interpretation (Scheme III). As the center at $\mathrm{C}-\delta$ determines the sense of the asymmetric induction, the established models ( $\mathbf{A}^{12 \mathrm{a}} \mathbf{B},{ }^{12 \mathrm{~b}}$ and $\mathbf{C}^{12 \mathrm{c}}$ in Scheme III) for acyclic 1,2-chirality transfer may be used. C- $\delta$ bears no heteroatom; consequently, the substituents in formulas $10 / 11$ may be assigned according to their inert volumes (i.e., $S=H, M=M e, L=$

Table II. Hiyama Reaction of Achiral Aldehydes with Bromides 13, 15, 19, 22, 24, and 27 (THF, $0^{\circ} \mathrm{C}$ )

| bromide | aldehyde | diasteromeric products | diastereomeric ratio | $\begin{gathered} \Delta \Delta G^{*}\left(0^{\circ} \mathrm{C}\right), \\ \mathrm{kcal} / \mathrm{mol} \end{gathered}$ | combined yield. \% |
| :---: | :---: | :---: | :---: | :---: | :---: |


| 13 | PhCHO |  |  | $\begin{gathered} 28: 29= \\ 83: 17 \end{gathered}$ | 0.86 | 68 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 15 | PhCHO |  |  <br> 31 | $\begin{gathered} 30: 31= \\ 82: 18 \end{gathered}$ | 0.82 | 80 |
| 19 | PhCHO |  |  | $\begin{gathered} 32: 33= \\ 93: 7 \end{gathered}$ | 1.40 | 68 |
| 22 | PhCHO |  |  | $\begin{gathered} 34: 35= \\ 91: 9 \end{gathered}$ | 1.26 | 84 |
| 24 | PhCHO |  |  | $\begin{gathered} 36: 37= \\ 96: 4 \end{gathered}$ | 1.72 | 75 |
| 24 | $\mathrm{H}_{2} \mathrm{C}_{13} \mathrm{CHO}$ |  |  | $\begin{aligned} & 38: 39= \\ & 96: 4 \end{aligned}$ | 1.72 | 60 |
| 27 | $\mathrm{H}_{2} \mathrm{C}_{13} \mathrm{CHO}$ |  |  | $\begin{aligned} & 40: 41= \\ & 96: 4 \end{aligned}$ | 1.72 | 60 |

$\mathrm{C}-\epsilon, \zeta, \eta) .{ }^{17}$ Additional stereocenters at $\mathrm{C}-\epsilon, \zeta$ apparently just increase the effective size of $L$. The major diastereomer is represented by formula 10 in all cases.

Models A-C are all based on the assumption of a Zimmer-man-Traxler ${ }^{18}$ chair geometry with an approximately antiperiplanar arrangement of L and the bulky $\mathrm{CrR}_{n}^{\prime}$ moiety. Model A only is in accord with the observed preference for diastereomer 10, so we have to discuss the advantages A may have over the alternatives B and C. It is easy to discard the allylic 1,3 -strain

[^2]model $\mathbf{C}$, because in our allylic bromides there is no cis substituent at $C \beta$, in contrast to the otherwise related ester enolates $47,{ }^{19}$ which in fact do add electrophiles according to model $\mathbf{C}$. Why is model $\mathbf{B}$ inferior to $\mathbf{A}$ ? The main reason may be the angle $\vartheta$ formed by the trajectory and the olefinic plane. According to the Zim-merman-Traxler model $\vartheta$ should be obtuse which is more in accord with model $\mathbf{A}$ (normal angle $\vartheta$ of ca. $110^{\circ}$ ) than with model B, for which angles $\vartheta$ of $\leqslant 90^{\circ}$ have been postulated. On comparing the steric crowding in both models, $\mathbf{A}$ is good with respect to "outside crowding" ( $\mathrm{R} \leftarrow \rightarrow \mathrm{S}$ ) and bad with respect to "inside
(19) (a) McGarvey, G. J.; Williams, J. M. J. Am. Chem. Soc. 1985, 107, 1435. (b) Bernhard. W.; Fleming, I.: Walerson. D. J. Chem. Soc., Chem. Commun. 1984, 28.


47
crowding" ( M syn to two "ring" bonds). The reverse is true of B. Inspection of molecular models reveals that in fact substituent R comes very close to $\mathbf{S}$ (in $\mathbf{A}$ ) and $\mathbf{M}$ (in B), respectively. Thus, the stereocontrolling factor in the addition obviously is to avoid outside crowding. Finally, Felkin-Anh boat conformations (D) also may be excluded: $\mathbf{D}$ would suffer from severe $R / R^{\prime}$ interactions and cannot compete with the chair (A).

In conclusion, the Hiyama addition of our chiral allylic bromides can stereochemically be described in terms of the Felkin-Anh model. Previously we have shown that the same model may also be applied in the addition of achiral allylic bromides to $\alpha$-chiral aldehydes $48 .{ }^{20}$ Thus, both components of the Hiyama reaction obey the same stereocontrolling principle! This result is somewhat surprising as the Felkin-Anh model normally is considered as characteristic of electron-deficient reaction sites.
Obviously, it is not the positive or negative polarization of the reaction site which decides on Felkin-Anh vs Houk transition states but the relative importance of "inside" and "outside" crowding and, hence, the angle $\vartheta$ of the trajectory. $\vartheta$, in turn, is determined by the overall geometry of the transition state and varies significantly with the individual reaction types. ${ }^{12 \mathrm{~b}}$

Double Stereodifferentiation (Tables III and IV). As mentioned above, $\alpha$-chiral aldehydes like ( $S$ )- and ( $R$ )-48 react with achiral allylic bromides like 49 under Felkin-Anh control to yield the anti-diol diastereomers $\mathbf{5 0}$ preferentially. The ratio of $\mathbf{5 0 : 5 1}$ varies between 89:11 and >99:1, depending on the nature of $\mathrm{R}^{2}$ and the

$+$


49
$\begin{array}{rr}\text { a } R^{1}=\text { THP } & R^{2}=M e, E, \\ b \quad R^{1}=\text { TBDMS } & n B u, P h\end{array}$


50


51

O-protective group $\mathrm{R}^{1}$. The highest selectivity was achieved for $\mathrm{R}^{1}=t-\mathrm{BuMe}_{2} \mathrm{Si}$ and $\mathrm{R}^{2}=n-\mathrm{C}_{4} \mathrm{H}_{9}$. Consequently, efficient double stereodifferentiation should be expected in additions of our chiral allylic bromides (e.g., 13, 19, 24, and 27) to aldehyde 48. As 48 is available in form of both enantiomers, we were sure that both the matched and the mismatched situation ${ }^{21}$ could be realized. From the stereoselectivities observed in either case, it should be possible to determine which component has the stronger stereodirecting influence.

The matched combinations are shown in Table III. For instance, adduct 52 has an all-syn arrangement of substituents at the centers $\mathrm{C} \delta, \mathrm{C} \gamma$, and $\mathrm{C} \beta^{\prime}$. It is obvious that the configuration at $C \beta^{\prime}$ also fits into the $\alpha^{\prime}, \beta^{\prime}$-anti-diol geometry demanded by the aldehyde.

The consonant stereochemical effects from both partners led to the exclusive ( $>97 \%$ ) formation of the adducts 52-54 in reasonable chemical yields (55-72\%). The configurational assignments are based on the experiments shown in Scheme IV. Thus, 52b was converted into the acetonide 55, in which the diol moiety is incorporated into a conformationally defined dioxolane ring. NOE experiments clearly show the cis location of H-4 and -5.

[^3]Scheme III. Transition State Models



B (Houk-Model) ${ }^{12 b}$


C (Allylic 1.3-Strain-Model) ${ }^{12 \mathrm{C}}$


D (Boat-Felkin-Anh-Model)
$\mathrm{R}=\mathrm{Cl}, \mathrm{Br}, \mathrm{THF}$
Table III. Double Stereodifferentiation: Matched Combinations
bromide aldehyde major diasteromeric product

53b was transformed into carbinol 56c via 56a-b. The mesylate 56d was unstable and cyclized to the tetrahydrofuran derivative 57 which was submitted to a single-crystal X-ray analysis (Figure 2). The spontaneous formation of 57 is surprising, as the central tetrahydrofuran ring has to accommodate two pseudo-axial substituents at $\mathrm{C}-4$ and $\mathrm{C}-5$, respectively. $\mathbf{5 4 b}$, like $\mathbf{5 2 b}$, was converted into the acetonide 58 and analyzed by NOE difference spectroscopy.
The mismatched combinations are the subject of Table IV. For instance, the addition of bromide 24 to aldehyde $(R)-59$ exclusively

${ }^{a}$ (a) $p$-TsOH catalyst, $\mathrm{Me}_{2} \mathrm{C}(\mathrm{OMe})_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} ; 22^{\circ} \mathrm{C}, 24 \mathrm{~h}, 84-98 \%$; (b) $\mathrm{O}_{3}$ and then $\mathrm{PPh}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2},-78$ to $22{ }^{\circ} \mathrm{C}, 85 \%$; (c) $\mathrm{LiAlH}_{4}, \mathrm{Et}_{2} \mathrm{O}$, $0-22{ }^{\circ} \mathrm{C}, 3 \mathrm{~h}, 95 \%$; (d) $\mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{Cl} / \mathrm{NEt}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0-22{ }^{\circ} \mathrm{C}, 24 \mathrm{~h}$, $71 \%$, (e) $p$-TsOH, MeOH and then $\mathrm{Me}_{2} \mathrm{C}(\mathrm{OMe})_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 22{ }^{\circ} \mathrm{C}$, $75 \%$.


Figure 2. Crystal structure of $\mathbf{5 7}$.
gave compound 60, with an all-syn arrangement at the stereocenters C-4-C-7. This means that the stereochemical course of the reaction has been dominated by the bromide, to which the aldehyde has succumbed. Similarly, bromide 27 reacts with ( $S$ )-59 to 62 with the same predominance of the bromide over the aldehyde. The configurations of $\mathbf{6 0}$ and 62 were proven by converting them into the natural product dihydrocanadensolide $((-)-61)$ and its enantiomer. ${ }^{22}$

Analogous results were obtained from the addition of 27 to $(S)-48 b$, which led to the all-syn products $63 \mathrm{a}, \mathrm{b}$. The configuration of the C-6,7-diol moiety in 63a was established by NOE experiments with the corresponding acetonide 64. So far, allylic bromides with three contiguous stereocenters have been employed. Remarkably, bromide 19, with two stereocenters, reacted with $(R)$-48b to give $\mathbf{6 5}$ with the same high diastereoselectivity. In contrast, bromide 13, with only one stereocenter, gave a mixture of three diastereomers with aldehyde ( $R$ )-48a. Obviously, not only the Felkin-Anh induction of the bromide but also the simple diastereoselection of the Hiyama addition itself have broken down! Finally, to test an extreme case, aldehyde 67a was tried. 67b had been shown to exhibit a very high ( $>95 \%$ ) anti-Felkin-Anh selectivity toward crotyl(II) chromium. ${ }^{23}$ This should lead to a mismatched combination of 67a with bromide 27. In fact, adduct 68 was obtained as a 2.5:1 mixture of diastereomers in low

[^4]Scheme V. ${ }^{\text {a }}$ Synthesis of Nephromopsinic Acid and Its Enantiomer

(-)-69



${ }^{a}$ (a) $p$-TsOH catalyst, $\mathrm{MeOH}, 22^{\circ} \mathrm{C}, 24 \mathrm{~h}, 84 / 85 \%$; (b) $\mathrm{Na} / \mathrm{NH}_{3}$, THF, $-40^{\circ} \mathrm{C}, 30 \mathrm{~min}, 92 / 94 \%$; (c) $\mathrm{H}_{5} \mathrm{IO}_{6}, \mathrm{Et}_{2} \mathrm{O}, 22^{\circ} \mathrm{C}, 1 \mathrm{~h}$ and then $\mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{MeOH}, 15 \mathrm{~min}, 73 \%$; (d) 5 equiv of $\mathrm{PCC}, \mathrm{CH}_{2} \mathrm{Cl}_{2}, 0-22^{\circ} \mathrm{C}, 2$ h, $92 \%$; (e) $\mathrm{RuCl}_{3}$ catalyst $/ \mathrm{NaIO}_{4}, \mathrm{CCl}_{4} / \mathrm{H}_{2} \mathrm{O} / \mathrm{CH}_{3} \mathrm{CN}, 22{ }^{\circ} \mathrm{C}, 2 \mathrm{~h}$, 62\%.
chemical yield (ca. 40\%). Sideproducts, resulting from the reduction and self-coupling of 27, were isolated in appreciable amounts. These results clearly mark the limits of the stereodominance of the allylic bromide: reliable stereocontrol in the sense of an all-syn arrangement at $\mathrm{C} \delta, \mathrm{C} \gamma$, and $\mathrm{C} \beta^{\prime}$ can only be expected for bromides with more than one chiral center and stereochemically "weak" aldehydes like 48 or 59 . Of course it remains to be clarified which structural elements in an aldehyde decide on its stereochemical "strength" or "weakness".

Synthesis of Nephromopsinic Acid (69) (Scheme V). Nephromopsinic acid ((-)-69) was isolated from the lichen species nephromopsis stracheyi by Asano and Azumi in 1935:24 its structure was assigned by van Tamelen and Huneck much later. ${ }^{25}$ As 69 has never been synthesized and its structure was only assigned on the basis of plausibility arguments, we chose this target molecule as a test case for our newly developed Hiyama methodology. To demonstrate the flexibility of the approach we prepared both enantiomers of 69 . The synthesis of ( - ). 69 (Scheme V) starts with adduct 38 , which was deprotected to give tetrol 70b. Degradation with periodic acid furnished lactol 71 and lactone 72 after oxidation with $\mathrm{PCC} . \mathrm{RuO}_{4}$ oxidation of the double bond delivered $(-) \cdot 69$, identical in every respect with an authentic sample of the natural product. Similarly, adduct 40 was converted into ( + ). 69 via 73a,b.

Epilogue. In conclusion, the addition of chiral allylic bromides to aldehydes under Hiyama conditions proceeds with high and reliable stereocontrol, the bromide acting as the stereodominant component. As both the aldehyde and the allylic bromide can easily be prepared from simple precursors, the reaction provides a rapid access to relatively complex adducts. Further studies of scope and limitation as well as applications to natural product synthesis are underway in our laboratory.

## Experimental Section

General Methods. NMR spectra were recorded on either a Bruker AM 270 SY or AC 250 instrument ( ${ }^{1} \mathrm{H}: 270 / 250 \mathrm{MHz}$, internal TMS;
(24) Asano, M.; Azumi, T. Ber. Dtsch. Chem. Ges. 1935, 68, 995.
(25) van Tamelen, E. E.; Rosenberg Bach, S. J. Am. Chem. Soc. 1958, 80, 3079. Huneck. S.; Follmann, G. Z. Naturforsch. 1967, 22, 666.

Table IV. Double Stereodifferentiation: Mismatched Combinations
comide
${ }^{13} \mathrm{C}: 68 / 63 \mathrm{MHz}$, internal $\mathrm{CDCl}_{3}$ ). IR spectra were recorded on a Perkin-Elmer 580 B infrared spectrophotometer. Mass spectra were measured at 80 eV on a Varian MAT 711 mass spectrometer. Optical rotations were obtained in $\mathrm{CHCl}_{3}$ with a Perkin-Elmer Model 241 polarimeter. HPLC separations were performed on Nucleosil 50 with particle sizes of $5 \mu \mathrm{~m}$ (analytical) and $7 \mu \mathrm{~m}$ (preparative), with UV and RI detection. Preparative column chromatography was performed on silica gel Merck $60,0.040-0.063 \mathrm{~mm}$. All reactions were carried oul in purified solvents under an argon atmosphere and were monitored on TLC plates (Merck 5554).
Starting Materials. $O$-Tetrahydropyranyl- and $\boldsymbol{O}$-(tert-butyldimethyl)silyl) lacetaldehydes $48 \mathrm{a} / \mathrm{b}$ were prepared from the corresponding O-protected lactates ${ }^{26}((S)$-series, ethyl ester; $(R)$-series, isobutyl ester). Aldehyde 67a, derived from D -( + )-ribono- 1,4 -lactone, was prepared as described in the literature for similar compounds. ${ }^{27}$ ( $R$ )-2,3-O-isopropylideneglyceraldehyde was prepared from D-mannitol. Tetradecanal was purchased from Fluka AG. ( $R$ )- and ( $S$ )-MOM-2-hydroxyhexanal 59 can be synthesized from ( $R$ )- and ( $S$ )-benzylglycidol (Aldrich) via a $\mathrm{Cu}(1)$-medialed epoxide opening with PrMgBr in the primary position

[^5](27) Jäger, V.; Häfele, B. Synthesis 1987, 801.
and conversion into the aldehydes by routine operations. ( $S$ ) -0 benzylglycidol could be prepared from D-mannitol according to a procedure described by Takano. ${ }^{28}$

Hiyama Reaction. General Procedure. In a typical experiment, $\mathrm{CrCl}_{3}$ $(4.28 \mathrm{~g}, 27 \mathrm{mmol})$ was suspended in THF ( 100 mL ). $\mathrm{LiAlH}_{4}(0.51 \mathrm{~g}$, 13.5 mmol ) was added in small portions under vigorous stirring at $0^{\circ} \mathrm{C}$. After the evolution of hydrogen had ceased, the mixture was stirred at $22^{\circ} \mathrm{C}$ for 30 min . The aldehyde ( 15 mmol ) and the allylic bromide ( 10 $\mathrm{mmol})$, boih dissolved in THF ( 20 mL ), were added at 0 to $-5^{\circ} \mathrm{C}$. After stirring the mixture for 36 h at this temperature, saturated aqueous sodium hydroxide ( 15 mL ) and anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}(20 \mathrm{~g})$ were added. The mixture was stirred for 20 min at $22^{\circ} \mathrm{C}$ and filtered over a pad of Celite/ $\mathrm{Na}_{2} \mathrm{SO}_{4}(7: 1)$. The filtrate was concentrated, and the residue was purified by column chromatography (hexane/ethyl acetate mixtures ( $3: 1$ to $10: 1$ )) to give the Hiyama adducts in $55-90 \%$ yield. The Hiyama adducts $28 / 29,38 / 39$, and 68 were separated by HPLC. Diastereomeric ratios were determined by ${ }^{1}$ H NMR and HPLC analysis or by weighing the diastereomers after separation.

Synthesis of Allylic Bromide 13 from 12a. Ethyl (4S)-5-(((tert-Bu-tyldiphenyl)silyl)oxy)-4-methyl-2 (E)-pentenoate (12b). $\mathrm{NaH}(0.36 \mathrm{~g}$, 15.0 mmol ) was suspended in THF ( 100 mL ), and triethyl phosphono-
(28) Takano, S.; Akiyama. M.; Ogasawara. K. Synthesis 1985, 503.
acetate ( $3.63 \mathrm{~g}, 16.2 \mathrm{mmol}$ ) in THF ( 50 mL ) was added dropwise at 0 ${ }^{\circ} \mathrm{C}$. After stirring for 2 h at room temperature, aldehyde $12 \mathrm{a}(4.05 \mathrm{~g}$, 12.3 mmol ) in THF ( 50 mL ) was added at $0^{\circ} \mathrm{C}$. After stirring for an additional 24 h at room temperature, the mixture was treated with $\mathrm{H}_{2} \mathrm{O}$ ( 20 mL ). THF was removed by evaporation under reduced pressure, and the residue was extracted with diethyl ether. Drying of the combined organic layers $\left(\mathrm{MgSO}_{4}\right)$ and evaporation of volatiles furnished the crude ester, which was purified by column chromatography (hexane/ethyl acetate (3:1). $12 \mathrm{~b}(4.65 \mathrm{~g}, 95 \%)$ was obtained as a colorless oil: $[\alpha]^{20} \mathrm{D}$ -9.7 ( $c 1.5, \mathrm{CHCl}_{3}$ ); IR (film) $\nu_{\max } 3070,2960,2930,2900,2860,1720$, $1650,1470,1460,1425,1390,1365,1305,1270,1240,1180,1150,1110$, $1035,1010,1000,985,825,805,740,700,690,615,505 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.06\left(\mathrm{~s}, 9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 1.07\left(\mathrm{~d}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.28$ (t, $3 \mathrm{H}, J=7 \mathrm{~Hz}$, ester- $\mathrm{CH}_{3}$ ), $2.54(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 3.58(\mathrm{~d}, 2 \mathrm{H}, \mathrm{H}-5)$, $4.19\left(\mathrm{q}, 2 \mathrm{H}, J=7 \mathrm{~Hz}\right.$, ester- $\left.\mathrm{CH}_{2}\right), 5.81(\mathrm{dd}, 1 \mathrm{H}, J=15.5,1 \mathrm{~Hz}, \mathrm{H}-2)$, 6.95 (dd, $1 \mathrm{H}, J=15.5,7 \mathrm{~Hz}, \mathrm{H}-3), 7.36(\mathrm{~m}, 6 \mathrm{H}$, aryl-H), $7.62(\mathrm{~m}, 4$ H, aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 14.25,15.57,19.26,26.58$, $26.83,39.08,60.11,67.59,121.10,127.64,129.64,133.61,135.59$, 151.24, 166.63; MS (EI $40^{\circ} \mathrm{C}$ ) m/e 340 (22.78), 339 (81.16), 227 (100), 199 (94.91), 183 (44.96). Anal. Calcd for $\mathrm{C}_{24} \mathrm{H}_{32} \mathrm{O}_{3} \mathrm{Si}: \mathrm{C}, 72.68 ; \mathrm{H}$, 8.13. Found: C, $72.40 ; \mathrm{H}, 8.18$.
(2S)-1-O-((tert-Butyldiphenyl)silyl)-2-methyl-3(E)-pentene-1,5-diol (12c). To a stirred solution of acrylic ester $12 \mathrm{~b}(4.50 \mathrm{~g}, 11.35 \mathrm{mmol})$ in diethyl ether ( 100 mL ), was added DIBAH ( $3.6 \mathrm{~g}, 17 \mathrm{~mL}, 25 \mathrm{mmol}$, 2.2 equiv as a 1.5 M solution in toluene) dropwise at $-40^{\circ} \mathrm{C}$. After stirring for 4 h at $-5^{\circ} \mathrm{C}$, water ( $2 \mathrm{~mL}, 111 \mathrm{mmol}, 10$ equiv) and $\mathrm{Na}_{2} \mathrm{SO}_{4}$ ( $16 \mathrm{~g}, 113 \mathrm{mmol}, 10$ equiv) were added, and the mixture was allowed to warm up to room temperature. After complete crystallization of the resulted precipitate, the mixture was filtered. Evaporation of volatiles and column chromatography (hexane/ethyl acetate (3:1)) provided 12c ( $3.21 \mathrm{~g}, 80 \%$ ): colorless oil; $[\alpha]^{20} \mathrm{D}-2.8\left(c 2.3, \mathrm{CHCl}_{3}\right.$ ); IR (film) $\nu_{\text {max }}$ $3350,3070,3050,2960,2930,2900,2860,2740,1470,1390,1360,1190$, $1110,1005,970,825,740,700,615,505 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( 250 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 1.04\left(\mathrm{~d}, 3 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.06\left(\mathrm{~s}, 9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 1.35$ ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{OH}), 2.40(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 3.49(\mathrm{dd}, 1 \mathrm{H}, J=10,6 \mathrm{~Hz}, \mathrm{H}-1)$, 3.53 (dd, $1 \mathrm{H}, J=10,7 \mathrm{~Hz}, \mathrm{H}-1), 4.05(\mathrm{mc}, 2 \mathrm{H}, \mathrm{H}-5), 5.67(\mathrm{mc}, 2 \mathrm{H}$, H-3, H-4), 7.39 (mc, 6 H , aryl-H), 7.65 (mc, 4 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR $\left(63 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 16.44,19.34,26.91,38.94,63.84,68.57,127.59$, $128.82,129.56,134.00,135.50,135.64$; MS (EI, $\left.40^{\circ} \mathrm{C}\right) \mathrm{m} / \mathrm{e} 199$ (100), 57 (91.28), 41 (89.9). Anal. Calcd for $\mathrm{C}_{22} \mathrm{H}_{30} \mathrm{O}_{2} \mathrm{Si}: \mathrm{C}, 74.53 ; \mathrm{H}, 8.53$. Found: C, 74.33; H, 8.63.
(2S)-5-Bromo-1-O-((tert-butyldiphenyl)silyl)-2-methyl-3(E)-penten-1-ol (13). Allylic alcohol $12 \mathrm{c}(7.50 \mathrm{~g}, 21.15 \mathrm{mmol})$ in diethyl ether ( 400 mL ) was treated dropwise under vigorous stirring with phosphorous tribromide ( $2.25 \mathrm{~g}, 8.31 \mathrm{mmol}, 0.39$ equiv) in ether ( 50 mL ) at $-25^{\circ} \mathrm{C}$. After stirring for an additional 2 h at $22^{\circ} \mathrm{C}$, the solution was poured into a concentrated aqueous $\mathrm{NaHCO}_{3}$ solution ( 200 mL ) at $0^{\circ} \mathrm{C}$. The organic layer was separated, and the aqueous phase was extracted with ether ( 3 times with 100 mL each). The combined etheric phases were dried $\left(\mathrm{MgSO}_{4}\right)$, and, after evaporation of volatiles, the chromatographed residue (hexane/ethyl acetate ( $3: 1$ )) gave $13(8.63 \mathrm{~g}, 98 \%)$ as a colorless oil: $[\alpha]^{20}{ }_{\mathrm{D}}+1.1\left(c\right.$ 1.1, $\left.\mathrm{CHCl}_{3}\right)$; IR (film) $\nu_{\max } 3070,3050,2960,2930$, $2900,2860,1470,1425,1390,1360,1205,1110,1090,1030,1010,700$, $615,505,490 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.01(\mathrm{~d}, 3 \mathrm{H}, J=$ $7.5 \mathrm{~Hz}, \mathrm{CH}_{3}$ ), $1.04\left(\mathrm{~s}, 9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 2.40(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-2), 3.51(\mathrm{~d}, 2$ $\mathrm{H}, \mathrm{J}=7 \mathrm{~Hz}, \mathrm{H}-5), 3.91(1,2 \mathrm{H}, \mathrm{H}-1), 5.68$ (mc, $2 \mathrm{H}, \mathrm{H}-3, \mathrm{H}-4), 7.36$ (m, 6 H , aryl-H), $7.62\left(\mathrm{~m}, 4 \mathrm{H}\right.$, aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 16.14,19.30,26.89,33.31,38.86,68.25,126.12,127.61,129.58,133.85$, 135.62, 138.84; MS (EI, $80^{\circ} \mathrm{C}$ ) $m / e 293$ (26.19), 263 (97.51), 261 (100), 135 (61.75), 91 (64.94), 57 (66.76). Anal. Calcd for $\mathrm{C}_{22} \mathrm{H}_{29} \mathrm{OSiBr}$ : $\mathrm{C}, 63.30 ; \mathrm{H}, 7.00$. Found: $\mathrm{C}, 63.27 ; \mathrm{H}, 6.77$. Similarly, bromides $15.19,22,24$. and 27 were prepared in 42-70\% overall yields from the corresponding aldehydes (see supplementary material).

Reactions of Achiral Aldehydes with the Allylic Bromides 13, 19, and 24. ( $1 R / S, 2 S / R, 3 S)-4-O$-( (tert - Butyldiphenyl)silyl)-3-methyl-1-phenyl-2-vinylbutane-1,4-diol (28/29). According to the general procedure described above, benzaldehyde ( $0.95 \mathrm{~g}, 8.95 \mathrm{mmol}, 1.5$ equiv) and $13(2.50 \mathrm{~g}, 5.99 \mathrm{mmol})$ were treated with a suspension of $\mathrm{CrCl}_{3}(2.56 \mathrm{~g}$, $16.2 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.31 \mathrm{~g}, 8.17 \mathrm{mmol})$ in THF $(100 \mathrm{~mL})$ to provide after column chromatography (hexane/ethyl acetate (5:1)) 28/29 ( $1.80 \mathrm{~g}, 68 \%$ ) as a colorless oil in a diastereomeric mixture (ratio 83:17). which was separated by HPLC ( $0.8 \%$ 2-propanol in hexane). 28: $[\alpha]^{20}{ }_{D}$ $+11.4\left(c 1.1, \mathrm{CHCl}_{3}\right)$; IR (film) $\nu_{\max } 3440,3070,3030,3000,2960,2930$, $2900,2860,1470,1460,1450,1425,1390,1360,1190,1110,1090,1030$, $1000,915,820,800,765,740,700,615,505 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR $(250 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 0.73\left(\mathrm{~d}, 3 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.02\left(\mathrm{~s}, 9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 1.58$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}-3$ ), $2.42(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 2.77$ (ddd, $1 \mathrm{H}, J=10,9.5 \mathrm{~Hz}, \mathrm{H}-2$ ), $3.38(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-4), 4.64(\mathrm{~d}, 1 \mathrm{H}, \mathrm{H}-1), 5.14(\mathrm{dd}, 1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ) 5.24 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.78 (ddd, 1 $\mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic- CH$), 7.20-7.67(\mathrm{~m}, 15 \mathrm{H}$, aryl -H$) ;{ }^{13} \mathrm{C}$

NMR (63 MHz, $\mathrm{CDCl}_{3}$ ) $\delta 11.94,19.17,26.81,35.76,52.95,66.93$, $74.44,120.11,126.96,127.54,128.26,129.46,129.50,133.62,133.67$. $135.46,135.50,135.55,142.66$; MS (EI, $90^{\circ} \mathrm{C}$ ) $\mathrm{m} / \mathrm{e} 198$ (100), 171 (47.09). Anal. Calcd for $\mathrm{C}_{29} \mathrm{H}_{36} \mathrm{O}_{2} \mathrm{Si} ; \mathrm{C}, 78.33 ; \mathrm{H}, 8.16$. Found: C , 78.03; H, 8.18.
(1S,2R,3S)-Isomer 29: ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.86$ (d, 3 $\left.\mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.05\left(\mathrm{~s}, 9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 1.92(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3), 2.29$ (ddd, $1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-2), 3.24(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 3.44(\mathrm{dd}, 1 \mathrm{H}, J$ $=11,5 \mathrm{~Hz}, \mathrm{H}-4), 3.59$ (dd, $1 \mathrm{H}, J=11,7.5 \mathrm{~Hz}, \mathrm{H}-4$ ), 4.85 (dd, 1 H , $J=17,2.5 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), $4.90(\mathrm{~d}, 1 \mathrm{H}, \mathrm{H}-1), 5.11(\mathrm{dd}, 1 \mathrm{H}, J=10$, 2.5 Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.85 (ddd, $J=17,10,10 \mathrm{~Hz}$, vinylic-CH), 7.34 and 7.65 (each mc, 15 H , aryl-H).
(3S,4S,5S/R,6R/S)-1,3-Di-O-benzyl-4-methyl-6-phenyl-5-vinyl-hexane-1,3,6-triol (32/33). According to the general procedure, benzaldehyde ( $2.81 \mathrm{~g}, 26.5 \mathrm{mmol}, 1.8$ equiv) and $19(5.93 \mathrm{~g}, 14.7 \mathrm{mmol}$ ) were treated with a suspension of $\mathrm{CrCl}_{3}(6.28 \mathrm{~g}, 39.7 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.75$ $\mathrm{g}, 19.8 \mathrm{mmol}$ ) in THF ( 200 mL ) to provide after column chromatography (hexane/ethyl acetate (3:1)) 32/33 (4.32 g, 68\%) (ratio 93:7): colorless oil. Major diastereomer 32: IR (film) $\nu_{\max } 3450,3065,3030$, $2970,2930,2870,1495,1450,1360,1205,1090,1025,1000,915,765$, $735,700 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.94(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}$, $\mathrm{CH}_{3}$ ), 1.76 (mc, $3 \mathrm{H}, \mathrm{H}-2, \mathrm{H}-4$ ), 2.48 (ddd, $1 \mathrm{H}, J=10,8,5 \mathrm{~Hz}, \mathrm{H}-5$ ), $2.48(\mathrm{~d}, 1 \mathrm{H}, J=2.5 \mathrm{~Hz}, \mathrm{OH}), 3.40(\mathrm{mc}, 3 \mathrm{H}, \mathrm{H}-1, \mathrm{H}-3), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=4.21, \delta_{\mathrm{B}}=4.27,2 \mathrm{H}, J_{\mathrm{AB}}=11.5 \mathrm{~Hz}\right.$, benzyl- $\left.\mathrm{CH}_{2}\right), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}\right.$ $=4.35, \delta_{\mathrm{B}}=4.41,2 \mathrm{H}, J_{\mathrm{AB}}=11.5 \mathrm{~Hz}$, benzyl $-\mathrm{CH}_{2}$ ), 4.62 (dd, $1 \mathrm{H}, J$ $=8,2.5 \mathrm{~Hz}, \mathrm{H}-6$ ), 5.12 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.21 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.82 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic-CH), $7.25\left(\mathrm{mc}, 15 \mathrm{H}\right.$, aryl-H); $\left.{ }^{13} \mathrm{C} \mathrm{NMR} \mathrm{(63} \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $12.18,30.93,35.83,53.37,66.96,71.75,72.87,73.97,79.79,119.55$, 126.80-128.27, 136.41, 138.47, $138.55,142.56$; MS (EI, $60^{\circ} \mathrm{C}$ ) m/e 107 (15.4), $91\left(100,\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}\right), 77\left(8.32,\left[\mathrm{C}_{6} \mathrm{H}_{5}\right]^{+}\right)$. Anal. Calcd for $\mathrm{C}_{29} \mathrm{H}_{34} \mathrm{O}_{3}: \mathrm{C}, 80.89 ; \mathrm{H}, 7.96$. Found: $\mathrm{C}, 80.60 ; \mathrm{H}, 7.94$.
( $2 R, 3 S, 4 S, 5 S / R, 6 R / S$ )-3-O-Benzyl-1,2-O-isopropylidene-4 methyl-6-phenyl-5-vinylhexane- $1,2,3,6$-tetrol $(36 / 37)$. According to the general procedure, benzaldehyde ( $0.76 \mathrm{~g}, 7.16 \mathrm{mmol}$ ) and $24(1.80 \mathrm{~g}$, 4.87 mmol ) were treated with a suspension of $\mathrm{CrCl}_{3}(2.0 \mathrm{~g}, 13 \mathrm{~mol})$ and $\mathrm{LiAlH}_{4}(0.25 \mathrm{~g}, 6.6 \mathrm{mmol})$ in THF ( 70 mL ) to give after column chromatography (hexane/ethyl acetate (3:1)) $36 / 37(1.45 \mathrm{~g}, 75 \%)$ as a colorless oil in a diastereomeric ratio of 96:4. 36: ${ }^{1} \mathrm{H} \mathrm{NMR}(250 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 0.90\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.30$ and 1.33 (each s, each 3 H , acetonide), $1.56(\mathrm{~m}, 1 \mathrm{H}, J=7.5,4.5 \mathrm{~Hz}, \mathrm{H}-4), 2.30(\mathrm{~d}, 1 \mathrm{H}, J=$ $2.5 \mathrm{~Hz}, \mathrm{OH}$ ), 2.76 (ddd, $1 \mathrm{H}, J=10,7.5,4.5 \mathrm{~Hz}, \mathrm{H}-5$ ), 3.56 (dd, 1 H , $J=7.5,4.5 \mathrm{~Hz}, \mathrm{H}-3) 3.72(\mathrm{dd}, 1 \mathrm{H}, J=9,8 \mathrm{~Hz}, \mathrm{H}-1), 3.82(\mathrm{dd}, 1 \mathrm{H}$, $J=9,7 \mathrm{~Hz}, \mathrm{H}-1), 4.14$ (ddd, $1 \mathrm{H}, J=8,7,4.5 \mathrm{~Hz}, \mathrm{H}-2$ ), AB system $\left(\delta_{\mathrm{A}}=4.48, \delta_{\mathrm{B}}=4.72,2 \mathrm{H}, J_{\mathrm{AB}}=11 \mathrm{~Hz}\right.$, benzyl- $\left.\mathrm{CH}_{2}\right), 4.62(\mathrm{dd}, 1 \mathrm{H}$, $J=7.5,2.5 \mathrm{~Hz}, \mathrm{H}-6), 5.00\left(\mathrm{dd}, 1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}\right.$, vinylic- $\mathrm{CH}_{2}$ ), 5.21 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.76 (ddd, $1 \mathrm{H}, J=17.5,10,10$ Hz , vinylic- CH ), $7.26(\mathrm{mc}, 10 \mathrm{H}$, aryl- H$) ;{ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 12.20,25.38,26.31,35.80,52.41,65.44,73.52,74.41,76.71,80.32$, 108.58, 119.97, 126.75-128.15, 135.88, 138.85, 142.45. Anal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{32} \mathrm{O}_{4}: \mathrm{C}, 75.72 ; \mathrm{H}, 8.13$. Found: $\mathrm{C}, 75.98 ; \mathrm{H}, 8.17$.
(2R,3S,4S,5S/R,6S/R)-3-O-Benzyl-1,2-O-isopropylidene-4-methyl-5-vinylnonadecane-1,2,3,6-tetrol (38/39). According to the general procedure, tetradecanal $(6.29 \mathrm{~g}, 29.6 \mathrm{mmol})$ and allylic bromide 24 $(7.29 \mathrm{~g}, 19.7 \mathrm{mmol})$ were treated with a suspension of $\mathrm{CrCl}_{3}(8.44 \mathrm{~g}, 53.3$ $\mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(1.01 \mathrm{~g}, 26.6 \mathrm{mmol})$ in THF $(300 \mathrm{~mL})$ to give after purification by column chromatography (hexane/ethyl acetate (10:1)) 38/39 ( $5.94 \mathrm{~g}, 60 \%$ ) (ratio 96:4): colorless oil. Major diastereomer 38: $[\alpha]^{20} \mathrm{D}+8.3$ ( $c 0.89, \mathrm{CHCl}_{3}$ ); IR (film) $\nu_{\text {max }} 3490,2920,2850,1450$, $1375,1365,1250,1205,1155,1140-970,910,850,730,695 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.88(\mathrm{t}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{H}-19$ ), 0.91 (d, 3 $\left.\mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.14-1.50(\mathrm{~m}, 24 \mathrm{H}, \mathrm{H}-7-18), 1.36$ and 1.40 (each s, each 3 H , acetonide), $1.54(\mathrm{~d}, 1 \mathrm{H}, \mathrm{OH}), 1.94(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4)$, 2.37 (ddd, $1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-5$ ), 3.68 (mc, $2 \mathrm{H}, \mathrm{H}-3, \mathrm{H}-6$ ), 3.92 (t, $1 \mathrm{H}, J=8,8 \mathrm{~Hz}, \mathrm{H}-1$ ), 4.03 (dd, $1 \mathrm{H}, J=8,6 \mathrm{~Hz}, \mathrm{H}-1$ ), 4.26 (ddd, $1 \mathrm{H}, J=8,6,5 \mathrm{~Hz}, \mathrm{H}-2), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=4.59, \delta_{\mathrm{B}}=4.80,2 \mathrm{H}, J_{\mathrm{AB}}\right.$ $=11 \mathrm{~Hz}$, benzyl $-\mathrm{CH}_{2}$ ), $5.05\left(\mathrm{dd}, 1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}\right.$, vinylic- $\mathrm{CH}_{2}$ ), 5.23 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.73 (ddd, $1 \mathrm{H}, J=17.5,10,10$ Hz , vinylic- CH ), 7.33 (mc, 5 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 12.88,14.06,22.44-35.91,25.38,25.53,50.99,66.07,71.41,73.45$, $76.53,80.20,108.66,119.02,127.44-128.27,136.39,138.82$; MS (EI, $\left.100{ }^{\circ} \mathrm{C}\right) \mathrm{m} / \mathrm{c} 91\left(100,\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}\right)$. Anal. Calcd for $\mathrm{C}_{32} \mathrm{H}_{94} \mathrm{O}_{4}: \mathrm{C}, 76.45$; $\mathrm{H}, 10.83$. Found: $\mathrm{C}, 76.18 ; \mathrm{H}, 10.74$.

Minor diastereomer 39 ( $4 \%$ of the combined yield, separated by HPLC (1\% 2-propanol in hexane)): ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.89$ $(\mathrm{t}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{H}-19), 1.06\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.18-1.49$ ( $\mathrm{m}, 24 \mathrm{H},[\mathrm{C}-7-\mathrm{C}-18]-\mathrm{CH}_{2}$ ), 1.34 and 1.38 (each s, each 3 H , acetonide), $1.56(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 2.15(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-4, \mathrm{H}-5), 3.69$ (dd, $1 \mathrm{H}, J=5.5,3$ $\mathrm{Hz}, \mathrm{H}-3), 3.80(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6), 3.84(\mathrm{dd}, 1 \mathrm{H}, J=8,8 \mathrm{~Hz}, \mathrm{H}-1), 4.04$ (dd, $1 \mathrm{H}, J=8,5.5 \mathrm{~Hz}, \mathrm{H}-1), 4.20$ (ddd, $1 \mathrm{H}, J=8,7.5,5 \mathrm{~Hz}, \mathrm{H}-2$ ),

AB system $\left(\delta_{\mathrm{A}}=4.56, \delta_{\mathrm{B}}=4.61,2 \mathrm{H}, J_{\mathrm{AB}}=10.5 \mathrm{~Hz}\right.$, benzyl- $\mathrm{CH}_{2}$ ), 5.08 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), $5.26(\mathrm{dd}, 1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ). 5.75 (ddd, $1 \mathrm{H} . J=17.5,10,10 \mathrm{~Hz}$, vinylic-CH), 7.32 (mc, 5 H , aryl-H).
(3S,4S,5S,6R)-4-Methyl-6-phenyl-5-vinylhexane-1,3,6-triol (44). The diastereomeric mixture of $32 / 33(1.50 \mathrm{~g}, 3.48 \mathrm{mmol})$ in THF ( 75 mL ) was added to ammonia ( 75 mL ) under vigorous stirring at $-40^{\circ} \mathrm{C}$. Sodium chips were added, until the solution became a deep blue color. After stirring for an additional 30 min , powdered $\mathrm{NH}_{4} \mathrm{Cl}$ was added, until the mixture became colorless. Warming up to $22^{\circ} \mathrm{C}$ under evaporation of the ammonia (ca. 3 h ), followed by filtration over a pad of Celite and concentration of the filtrate furnished after purification by column chromatography (ethyl acetate) and recrystallization from diisopropyl ether $44(0.79 \mathrm{~g}, 91 \%)$ in a diastereomeric purity of $>99 \%$. Colorless needles; $\mathrm{mp} 99^{\circ} \mathrm{C} ;[\alpha]^{20} \mathrm{D}+47.5\left(c 1.2, \mathrm{CHCl}_{3}\right)$; IR (KBr) $\nu_{\max } 3200$, 3070, 3030, 2970, 2880, 1740, 1490, 1450, 1420, 1375, 1350, 1325, 1290, $1230,1180,1135,1080,1050,1020,990,935,910,860,820,760,720$, $700,680,640,580,525 \mathrm{~cm}^{-1}:{ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.02(\mathrm{~d}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), 1.52 (mc, $2 \mathrm{H}, \mathrm{H}-2$ ), $1.72(\mathrm{ddq}, 1 \mathrm{H}, J=5 \mathrm{~Hz}, \mathrm{H}-4), 2.37$ (ddd, $1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-5), 2.97(\mathrm{t}, 1 \mathrm{H}, J=5 \mathrm{~Hz}, \mathrm{OH}), 3.49$ (s, $1 \mathrm{H}, \mathrm{OH}), 3.57(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 3.94(\mathrm{mc}, 2 \mathrm{H}, \mathrm{H}-1), 3.99(\mathrm{dt}, 1 \mathrm{H}, \mathrm{H}-3)$, 4.82 (d, $1 \mathrm{H}, J=5 \mathrm{~Hz}, \mathrm{H}-6), 4.94$ (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic$\mathrm{CH}_{2}$ ), 5.15 (dd, $1 \mathrm{H} . J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.98 (ddd, $1 \mathrm{H}, J=$ 17.5, $10,10 \mathrm{~Hz}$, vinylic-CH), 7.27 (mc, 5 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR (63 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 11.02,35.78,40.51,55.44,62.04,72.74,73.26,119.02$, $126.45,127.33,128.14,136.51,142.86$; MS (EI, FAB) $m / e 501$ (1.76, $\left.[2 \mathrm{M}+\mathrm{H}]^{+}\right), 251\left(12.63,[\mathrm{M}+\mathrm{H}]^{+}\right), 233(100), 159$ (22.3), 131 (24.81), 107 (26.19), 105 (27.26), 93 (20.28), 91 (24.9). Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{22} \mathrm{O}_{3}$ : C. $71.97 ; \mathrm{H}, 8.86$. Found: $\mathrm{C}, 71.83 ; 8.72$.

Double Stereodifferentation: Matched Combinations. ( $\left.\mathbf{2 S}, \mathbf{3 S}, 4 R, 5 S, 2^{\prime} R / S\right)-1-O$-( $($ tert-Butyldiphenyl) silyl)-2-methyl-5-O. tetrahydropyranyl-3-vinylhexane-1,4,5-triol (52a). According to the general procedure described above, $(S)-48 \mathrm{a}(1.42 \mathrm{~g}, 8.98 \mathrm{mmol})$ and 13 $(2.50 \mathrm{~g}, 5.99 \mathrm{mmol})$ in THF ( 30 mL each) were treated with a suspension of $\mathrm{CrCl}_{3}(2.65 \mathrm{~g}, 16.7 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.31 \mathrm{~g}, 8.17 \mathrm{mmol})$ in THF $(100 \mathrm{~mL})$ to give after column chromatography (hexane/ethyl acetate (3:1)) 52a ( $1.84 \mathrm{~g}, 62 \%$ ) as a colorless oil: IR (film) $v_{\max } 3470,3070$, $3050,2940,2860,1470,1425,1390,1360,1260,1200,1185,1110,1075$, $1025,995,935,915,870,820,740,700,615,505 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR (250 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.92\left(\mathrm{~d}, 6 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-2-\mathrm{CH}_{3}\right), 1.06(\mathrm{~s}, 18 \mathrm{H}$, $\left.t-\mathrm{BuCH}_{3}\right), 1.17(\mathrm{~d}, 3 \mathrm{H}, J=6 \mathrm{~Hz}, \mathrm{H}-6), 1.31(\mathrm{~d}, 3 \mathrm{H}, J=6 \mathrm{~Hz}, \mathrm{H}-6)$, 1.44-1.93 (m. 14 H, THP-H-3', $4^{\prime} .5^{\prime}, \mathrm{H}-2$ ), 2.45 (mc, $2 \mathrm{H}, \mathrm{OH}, \mathrm{H}-3$ ), 2.56 (mc, $2 \mathrm{H}, \mathrm{OH}, \mathrm{H}-3$ ), 3.48 (mc, $6 \mathrm{H}, \mathrm{H}-5, \mathrm{H}-1$ ), 3.82 (m, $2 \mathrm{H}, \mathrm{H}-4$ ), 3.92 (m, 4 H, THP-H-6'), 4.69 (m, 2 H, THP-H-2'), 5.16 (m, $4 \mathrm{H}, \mathrm{J}$ $=17.5,10,2 \mathrm{~Hz}$, vinylic $\left.-\mathrm{CH}_{2}\right), 6.20(\mathrm{~m}, 2 \mathrm{H}, J=17.5,10 \mathrm{~Hz}$, vinyl-ic-CH), $7.39\left(\mathrm{~m}, 12 \mathrm{H}\right.$, aryl-H), $7.66\left(\mathrm{~m}, 8 \mathrm{H}\right.$, aryl-H); ${ }^{13} \mathrm{C}$ NMR ( 63 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 12.10,12.74,13.11,15.86,19.17,19.84,25.44,26.83$, $31.16,35.87 .36 .03,46.67,47.68,62.78,62.91,66.90,67.05,72.73,72.89$, $73.68,74.87,96.86,98.65,118.35,118.51,127.65,129.65,133.52$, 135.52, $135.61,135.93,136.37$; MS (EI, $180^{\circ} \mathrm{C}$ ) $m / e 337$ (30.39), 199 (79.72), 85 (100). Anal. Calcd for $\mathrm{C}_{30} \mathrm{H}_{44} \mathrm{O}_{4} \mathrm{Si}: \mathrm{C}, 72.54 ; \mathrm{H}, 8.92$. Found: C, 72.28; H, 8.62.
( $2 S, 3 S, 4 R, 5 S$ )-1-O ((tert -Butyldiphenyl) silyl)-2-methyl-3-vinyl-hexane-1,4,5-triol (52b). To a stirred solution of $52 \mathrm{a}(1.43 \mathrm{~g}, 2.88 \mathrm{mmol})$ in $\mathrm{MeOH}(200 \mathrm{~mL})$ was added PPTS until the pH was well below 3.5. After stirring for 24 h at $22^{\circ} \mathrm{C}$, the mixture was neutralized with powdered $\mathrm{NaHCO}_{3}$ and concentrated. Dilution with ether ( 100 mL ), drying ( $\mathrm{MgSO}_{4}$ ), and filtration furnished after purification by column chromatography (hexane/ethyl acetate (1:1)) 52 b ( $0.64 \mathrm{~g}, 54 \%$ ): white crystals; $\mathrm{mp} 40{ }^{\circ} \mathrm{C} ;[\alpha]^{20} \mathrm{D}-1.7\left(c 1.05, \mathrm{CHCl}_{3}\right)$; IR (KBr) $\nu_{\text {max }} 3420$, $3070,3050,2960,2930,2900,2860,1470,1425,1390,1360,1110,1085$, 1060, 1005, $1000,985,920,825,810,740,700,615 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H}$ NMR (250 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.88\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-2-\mathrm{CH}_{3}\right), 1.07(\mathrm{~s}, 9 \mathrm{H}, t \cdot \mathrm{Bu}-$ $\mathrm{CH}_{3}$ ), 1.24 (d. $3 \mathrm{H} . J=6 \mathrm{~Hz} . \mathrm{H}-6$ ). 1.84 (m. $1 \mathrm{H}, \mathrm{H}-2$ ), 2.07 (d, 1 H , $J=6 \mathrm{~Hz}, \mathrm{OH}), 2.40(\mathrm{ddd}, 1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-3), 2.60(\mathrm{~d}, 1 \mathrm{H}$, $J=4 \mathrm{~Hz}, \mathrm{OH}) .3 .50(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-1), 3.71(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 3.82(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{H}-4$ ), 5.14 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.20 (dd, $1 \mathrm{H}, J=$ $10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.80 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$. vinylic- CH ), $7.38\left(\mathrm{mc}, 6 \mathrm{H}\right.$, aryl-H). 7.63 (mc, 4 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR ( 63 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 13.32,17.75,19.13,26.78,35.91,48.70,67.19,68.53,74.17$, $118.92,127.67,129.71,133.28,135.48,135.59,136.51$; MS (EI, $140^{\circ} \mathrm{C}$ ) $m / e 199$ (100), 139 (23.97), 57 (50.95), 55 (41.04). Anal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{3} \mathrm{Si}: \mathrm{C}, 72.77 ; \mathrm{H}, 8.79$. Found: $\mathrm{C}, 72.31 ; \mathrm{H}, 8.38$.
(2R,3S,4S,5S,6R,7S)-3-O-Benzyl-7-O-((tert-Butyldmethyl)si-lyl)-1,2-O-Isopropylidene-4-methyl-5-vinyloctane-1,2,3,6,7-pentol (53a). According to the general procedure, $24(3.47 \mathrm{~g}, 9.40 \mathrm{mmol})$ and $(S)-48 \mathrm{~b}$ ( $1.98 \mathrm{~g}, 10.58 \mathrm{mmol}$ ) in THF ( 40 mL each) were treated with a suspension of $\mathrm{CrCl}_{3}(4.0 \mathrm{~g}, 25.3 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.48 \mathrm{~g}, 12.7 \mathrm{mmol})$ in THF ( 100 mL ) 10 give afier column chromatography (hexane/ethyl acetate (5:1)) 53a ( $3.24 \mathrm{~g}, 72 \%$ ); colorless oil: $[\alpha]^{20} \mathrm{D}+10.8$ (c 2.4,
$\mathrm{CHCl}_{3}$ ); IR (film) $\nu_{\max } 3570,3035,3015,2980,2950,2930,2885,2860$, $1495,1469,1460,1450,1420,1375,1367,1348,1290,1252,1210,1152$, $1123,1075,1025,1004,967,938,912,872,833,775,732,695 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.07$ and 0.11 (each $s$, each $3 \mathrm{H}, \mathrm{Si}-\mathrm{CH}_{3}$ ), $0.90\left(\mathrm{~d}, 3 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 0.93\left(\mathrm{~s}, 9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 1.08(\mathrm{~d}$, $3 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{H}-8$ ), 1.36 and 1.42 (each s, each 3 H , acetonide), 1.84 (m, 1 H, H-4), $2.36(\mathrm{~d}, 1 \mathrm{H}, \mathrm{OH}), 2.61(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 2.60-4.30(\mathrm{~m}$, $6 \mathrm{H}, \mathrm{H}-1,-2,-3,-6,-7), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=4.59, \delta_{\mathrm{B}}=4.91,2 \mathrm{H}, J_{\mathrm{AB}}=\right.$ 12 Hz , benzyl $-\mathrm{CH}_{2}$ ), $5.06\left(\mathrm{dd}, 1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}\right.$, vinylic- $\mathrm{CH}_{2}$ ), 5.22 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.80 (ddd, $1 \mathrm{H}, J=17.5,10,10$ Hz , vinylic-CH), $7.32(\mathrm{mc}, 5 \mathrm{H}$, aryl- H$) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(63 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta-4.82,-4.28,12.41,17.01,17.97,25.39,25.80,26.40,36.59,46.22$, $65.05,69.75,74.11,74.83,77.13,79.57,108.54,118.49,127.34,127.41$, 128.22, 136.14, 138.91; MS (CI, $\left.50^{\circ} \mathrm{C}\right) \mathrm{m} / e 478\left(2,[\mathrm{M}]^{+}\right), 463$ (21, $\left.\left[\mathrm{M}-\mathrm{CH}_{3}\right]^{+}\right), 363(83), 313(68), 91\left(100,\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}\right)$. Anal. Calcd for $\mathrm{C}_{27} \mathrm{H}_{46} \mathrm{O}_{5} \mathrm{Si}: \mathrm{C}, 67.74 ; \mathrm{H}, 9.69$. Found: $\mathrm{C}, 67.99 ; \mathrm{H}, 9.70$.
(2R,3S,4S,5S,6R,7S)-3-O-Benzyl-1,2-O-isopropylidene-4-methyl5 -vinyloctane-1,2,3,6,7-pentol (53b). To a stirred solution of 53 a ( 10 g , 21 mmol ) in THF ( 300 mL ) was added $\mathrm{Bu}_{4} \mathrm{NF}$ (trihydrate) ( $10.3 \mathrm{~g}, 33$ mmol ) in THF ( 25 mL ) dropwise at $22^{\circ} \mathrm{C}$, and stirring was continued for 1 h . Water ( 100 mL ) was added, and the mixture was concentrated. Dilution and extraction with ether ( 200 mL each), followed by drying ( $\mathrm{MgSO}_{4}$ ) of the combined organic phases, furnished after concentration and purification of the residue by column chromatography (hexane/ethyl acetate (1:1)) 53b ( $6.9 \mathrm{~g}, 90 \%$ ): colorless oil; $[\alpha]^{20} \mathrm{D}+33.7$ (c 2.5, $\mathrm{CHCl}_{3}$ ); IR (film) $\nu_{\text {max }} 3650-3150,3075,3035,2980,2940,2890$, 2650-2400, 1640, 1500, 1355, 1420, 1378, 1368, 1350, 1325, 1295, 1248, $1212,1157,1120,1060,1028,1005,980,920,860,790,760,735,700$ $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.96\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right)$, 1.19 (d, $3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{H}-8$ ), 1.39 and 1.45 (each s, each 3 H , acetonide), $1.78(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 2.50(\mathrm{~s}, 2 \mathrm{H}, \mathrm{OH}), 2.59$ (ddd, $1 \mathrm{H}, J=10$, $5,5 \mathrm{~Hz}, \mathrm{H}-5), 3.52-4.34(\mathrm{~m}, 6 \mathrm{H}, \mathrm{H}-1,-2,-3,-6,-7)$, AB system ( $\delta_{\mathrm{A}}=$ $4.58, \delta_{\mathrm{B}}=4.86,2 \mathrm{H}, J_{\mathrm{AB}}=11.5 \mathrm{~Hz}$, benzyl $\left.-\mathrm{CH}_{2}\right), 5.09(\mathrm{dd}, 1 \mathrm{H}, J=$ $17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.25 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic $-\mathrm{CH}_{2}$ ), 5.81 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic-CH), $7.33\left(\mathrm{~m}, 5 \mathrm{H}\right.$, aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 15.70,18.93,25.45,26.50,37.70,48.12$, $66.49,68.53,74.06,78.33,82.03,109.50,118.20,127.20,127.50,128.23$, 136.29, 138.37; MS (CI, $160 \mathrm{eV}, 130^{\circ} \mathrm{C}$ ) m/e 365 (24), 289 (30), 199 (35), 181 (30), 155 (60), $91\left(56,\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}\right)$. Anal. Calcd for $\mathrm{C}_{21} \mathrm{H}_{32} \mathrm{O}_{5}$ : C, $69.20 ; \mathrm{H}, 8.85$. Found: C, $68.88 ; \mathrm{H}, 8.88$.
( $2 R, 3 R, 4 R, 5 R, 6 S, 7 R$ )-3-O-Benzyl-7-O-((tert -butyldimethyl) si-lyl)-1,2-O-isopropylidene-4-methyl-5-vinyloctane-1,2,3,6,7-pentol (54a). According to the general procedure, allylic bromide 27 ( $2.95 \mathrm{~g}, 7.99$ $\mathrm{mmol})$ and $(R)-48 \mathrm{~b}(2.13 \mathrm{~g}, 11.37 \mathrm{mmol})$ were treated with a suspension of $\mathrm{CrCl}_{3}(3.42 \mathrm{~g}, 21.6 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.41 \mathrm{~g}, 10.8 \mathrm{mmol})$ in THF $(100 \mathrm{~mL})$ to give after column chromatography (hexane/ethyl acetate (5:1)) 54a ( $2.10 \mathrm{~g}, 55 \%$ ): colorless oil; $[\alpha]^{20} \mathrm{D}+24.6\left(c 1.30, \mathrm{CHCl}_{3}\right)$; IR and MS data see 53a; ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.08$ ( $\mathrm{s}, 3 \mathrm{H}$, $\left.\mathrm{SiCH}_{3}\right), 0.10\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{SiCH}_{3}\right), 0.89\left(\mathrm{~s}, 9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 0.95(\mathrm{~d}, 3 \mathrm{H}, J$ $\left.=7.5 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.11(\mathrm{~d}, 3 \mathrm{H}, J=6.25 \mathrm{~Hz}, \mathrm{H}-8), 1.38$ and 1.46 (each s, each 3 H , acetonide), $1.81-1.94$ (m,1 H,H-4), 2.60 (ddd, 1 H , $J=8,7,6 \mathrm{~Hz}, \mathrm{H}-5), 2.84(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 3.33(\mathrm{dd}, 1 \mathrm{H}, J=6,5 \mathrm{~Hz}$, $\mathrm{H}-3$ ), 3.58 (dd, $1 \mathrm{H}, J=8,6 \mathrm{~Hz}, \mathrm{H}-6), 3.72(\mathrm{dd}, 1 \mathrm{H}, J=7.5,7 \mathrm{~Hz}$, $\mathrm{H}-1), 3.84(\mathrm{dq}, 1 \mathrm{H}, J=8,6.25 \mathrm{~Hz}, \mathrm{H}-7), 4.03(\mathrm{dd}, 1 \mathrm{H}, J=7,7.5 \mathrm{~Hz}$, $\mathrm{H}-1), 4.31$ (ddd, $1 \mathrm{H}, J=7.5,7,6 \mathrm{~Hz}, \mathrm{H}-2$ ), AB system $\left(\delta_{\mathrm{A}}=4.64, \delta_{\mathrm{B}}\right.$ $=4.81,2 \mathrm{H}, J_{\mathrm{AB}}=12 \mathrm{~Hz}$, benzyl- $\left.\mathrm{CH}_{2}\right), 5.10(\mathrm{dd}, 1 \mathrm{H}, J=16,2.5 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.25 (dd, $1 \mathrm{H}, J=10,2.5 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.89 (ddd, $1 \mathrm{H}, J=16,10,8 \mathrm{~Hz}$, vinylic-CH), $7.34(\mathrm{mc}, 5 \mathrm{H}$, aryl- H$) ;{ }^{13} \mathrm{C}$ NMR $\left(63 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-4.64,-4.14,13.98,17.95 .25 .50,25.93,26.57$, $37.93,46.65,66.58,70.12,74.27,75.16,78.06,82.64,108.94,118.18$, $127.49,127.78,128.29,136.99,138.90 ;$ HRMS Calcd for $\mathrm{C}_{27} \mathrm{H}_{46} \mathrm{O}_{5} \mathrm{Si}$ $\left([\mathrm{M}]^{+}\right) 478.31145$, found 478.31143.
( $2 R, 3 R, 4 R, 5 R, 6 S, 7 R$ )-3-O-Benzyl-1,2-O-isopropylidene-4-methyl5 -vinyloctane-1,2,3,6,7-pentol (54b). According to the procedure described above for the desilylation of $53 \mathrm{a}, 54 \mathrm{a}(2.0 \mathrm{~g}, 4.18 \mathrm{mmol})$ was treated with $\mathrm{Bu}_{4} \mathrm{NF}$ (trihydrate) $(2.06 \mathrm{~g}, 6.5 \mathrm{mmol})$ in THF $(50 \mathrm{~mL})$. Usual workup gave after purification by column chromatography (hexane/ethyl acetate (1:1)) $54 \mathrm{~b}\left(1.37 \mathrm{~g}, 90 \%\right.$ ) as a colorless oil; $[\alpha]^{20} \mathrm{D}+26.7$ (c $1.0, \mathrm{CHCl}_{3}$ ); IR and MS data see 53 b , ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.97\left(\mathrm{~d}, 3 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.18(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{H}-8), 1.37$ and 1.46 (each s, each 3 H , acetonide), $1.84-1.96(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 2.17$ ( $\mathrm{s}, 2 \mathrm{H}, \mathrm{OH}$ ), 2.56 (ddd, $1 \mathrm{H}, J=8.5,6.5,6.5 \mathrm{~Hz}, \mathrm{H}-5$ ), 3.31 (dd, 1 H , $J=7,5.5 \mathrm{~Hz}, \mathrm{H}-3$ ), 3.65 (dd, $1 \mathrm{H}, J=6.5,5.5 \mathrm{~Hz}, \mathrm{H}-6$ ), 3.71 (dd, 1 H. $J=8,8 \mathrm{~Hz}, \mathrm{H} \cdot 1$ ), 3.75-3.85 (m, 1 H, H-7), 4.03 (dd, $1 \mathrm{H}, J=8$, $6.5 \mathrm{~Hz}, \mathrm{H}-1$ ), 4.31 (ddd, $1 \mathrm{H}, J=8,7,6.5 \mathrm{~Hz}, \mathrm{H}-2$ ), AB system ( $\delta_{\mathrm{A}}=$ 4.61. $\delta_{B}=4.81,2 \mathrm{H}, J_{A B}=12 \mathrm{~Hz}$, benzyl- $\left.\mathrm{CH}_{2}\right), 5.11(\mathrm{dd}, 1 \mathrm{H}, J=16.5$, 2 Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.27 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.90 (ddd, $1 \mathrm{H}, J=16.5,10,8.5 \mathrm{~Hz}$, vinylic- CH ), $7.34(\mathrm{mc}, 5 \mathrm{H}$, aryl -H$){ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.79,17.53,25.38,26.44,37.19,47.63$, $66.29,68.71,73.90 .74 .20,77.76,82.28,108.84,119.25,127.54,127.66$,
128.28, 136.84, 138.39; HRMS calcd for $\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{O}_{5}\left(\left[\mathrm{M}-\mathrm{CH}_{3}\right]^{+}\right)$ 349.20150 , found 349.20156 .
( $\mathbf{2 S}, \mathbf{3 S}, \mathbf{4 R}, 5 S$ )-1-O (( tert -Butyldiphenyl) silyl)-4,5-O-iso-propylidene-2-methyl-3-vinylhexane-1,4,5-triol (55). To a stirred solution of diol $52 \mathrm{~b}(0.41 \mathrm{~g}, 0.99 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ was added PPTS until the pH became about 4 . DMP $(0.16 \mathrm{~g}, 1.5 \mathrm{mmol})$ was added, and the solution continued to be stirred for 1 h at $22^{\circ} \mathrm{C}$. The mixture was poured into a saturated aqueous $\mathrm{NaHCO}_{3}$ solution ( 50 mL ), and the aqueous phase was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure, and the residue was purified by column chromatography (hexane/ethyl acetate (3:1)) to give $55(0.44 \mathrm{~g}, 98 \%)$ : colorless oil; $[\alpha]^{20} \mathrm{D}-25.1\left(c 2.5, \mathrm{CHCl}_{3}\right)$; ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.77\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.05(\mathrm{~s}$, $\left.9 \mathrm{H}, t-\mathrm{BuCH}_{3}\right), 1.20(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{H}-6), 1.32$ and 1.44 (each s , each 3 H , acetonide), 1.74 (m,1 H, H-2), 2.73 (m, 1 H, H-3), 3.40 (dd, $1 \mathrm{H}, J=10,6.5 \mathrm{~Hz}, \mathrm{H}-1), 3.48(\mathrm{dd}, 1 \mathrm{H}, J=10,8.5 \mathrm{~Hz}, \mathrm{H}-1), 4.18$ (mc, $2 \mathrm{H}, \mathrm{H}-4, \mathrm{H}-5$ ), 5.08 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.14 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.69 (ddd, $1 \mathrm{H}, J=17.5,10,10$ Hz , vinylic- CH$), 7.36\left(\mathrm{~m}, 6 \mathrm{H}\right.$, aryl-H), $7.68(\mathrm{~m}, 4 \mathrm{H}$, aryl- H$)$; ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 11.50,16.23,19.18,25.98,26.85,28.52$, $36.63,42.85 .66 .52,73.85,78.50,107.39,117.92,127.60,129.56,129.60$, $133.75,135.32,135.50,135.59$; MS (EI, $90^{\circ} \mathrm{C}$ ) m/e 199 (100), 183 (24), 115 (53.48), $69(70.25), 59(33.62), 43$ (67.58). Anal. Calcd for $\mathrm{C}_{28} \mathrm{H}_{40} \mathrm{O}_{3} \mathrm{Si}: \mathrm{C}, 74.29 ; \mathrm{H}, 8.91$. Found: C, $74.21 ; \mathrm{H}, 9.11$
( $2 R, 3 S, 4 S, 5 S, 6 R, 7 S$ )-3- $O$-Benzyl-1,2:6,7-di- $O$-isopropylidene-4-methyl-5-vinyloctane-1,2,3,6,7-pentol (56a). Ketalization of 53b ( 8.0 g , 22 mmol ) was accomplished with DMP ( $3.5 \mathrm{~g}, 34 \mathrm{mmol}$ ) as described for the preparation of 55 to give after purification by column chromatography (hexane/ethyl acetate (5:1)) $56 \mathrm{a}(7.8 \mathrm{~g}, 88 \%$ ) as a colorless oil: $[\alpha]{ }^{20} \mathrm{D}-27.4\left(c 1.8, \mathrm{CHCl}_{3}\right)$; IR (film) $\nu_{\max } 3070,3030,2980,2935,2885$, $1495,1450,1380.1350,1300,1240,1215,1170,1160,1119,1065,1038$, $1028,1008,970,931,915,870,860,850,790,750,735,700 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.90\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.13$ (d, 3 $\mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{H}-8$ ) $, 1.30,1.35,1.39$, and 1.42 (each s, each 3 H , acetonides), 1.45 (m, 1 H, H-4), $2.79(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 3.65(\mathrm{dd}, 1 \mathrm{H}, J=9$. $3 \mathrm{~Hz}, \mathrm{H}-6), 3.95(\mathrm{dd}, 1 \mathrm{H}, J=9,7.5 \mathrm{~Hz}, \mathrm{H}-1), 3.99(\mathrm{dd}, 1 \mathrm{H}, J=9$, $9 \mathrm{~Hz}, \mathrm{H}-1), 4.10(\mathrm{dd}, 1 \mathrm{H}, J=9,6 \mathrm{~Hz}, \mathrm{H}-3), 4.18$ (ddd, $1 \mathrm{H}, J=9$, $7.5 \mathrm{~Hz}, \mathrm{H}-2), 4.28(\mathrm{dq}, 1 \mathrm{H}, J=7.3 \mathrm{~Hz}, \mathrm{H}-7), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=4.60\right.$, $\delta_{\mathrm{B}}=4.98,2 \mathrm{H}, J_{\mathrm{AB}}=11.5 \mathrm{~Hz}$, benzyl- $\left.\mathrm{CH}_{2}\right), 5.00(\mathrm{dd}, 1 \mathrm{H}, J=17,2$ Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.20 (dd. $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.78 (ddd, $1 \mathrm{H}, J=17,10,9 \mathrm{~Hz}$, vinylic-CH), $7.30\left(\mathrm{mc}, 5 \mathrm{H}\right.$, aryl-H); ${ }^{13} \mathrm{C}$ NMR $\left(63 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 11.73,15.68,25.16,25.53,26.08,27.96,37.70$, $42.91,64.12,73.71,74.18,77.09,78.29,78.82,107.01,108.27,117.99$, $127.05,127.13,127.93 .135 .48,138.60$ : MS (CI, $20^{\circ} \mathrm{C}$ ) $\mathrm{m} / \mathrm{e} 405$ (12.4, $\left.[\mathrm{MH}]^{+}\right), 289(100), 239(34.3), 91\left(75.75\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}\right)$; HRMS calcd for $\mathrm{C}_{23} \mathrm{H}_{33} \mathrm{O}_{5}\left(\left[\mathrm{M}-\mathrm{CH}_{3}\right]^{+}\right) 389.2328$, found 389.2329 .
(2'R,2"S,3S,4S,5R)-4-(Benzyloxy)-2-(( $2^{\prime}, 2^{\prime \prime}$-isopropylidenedioxy)-propyl)-3-methyl-5,6-(isopropylidenedioxy) hexanal (56b). 56 a ( 5.0 g , 12.4 mmol ) was dissolved in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(300 \mathrm{~mL})$, and at $-78^{\circ} \mathrm{C}$ ozone was passed through the solution until it became slightly blue. Excess ozone was removed with a stream of nitrogen (solution became colorless), and $\mathrm{PPh}_{3}(13 \mathrm{~g}, 50 \mathrm{mmol})$ was added. The mixture was warmed up to 22 ${ }^{\circ} \mathrm{C}$ during 5 h , followed by evaporation of the solvent under reduced pressure and dilution of the residue with ether ( 200 mL ). The precipitated $\mathrm{PPh}_{3} \mathrm{O}$ was separated by filtration, and the filtrate was concentrated. Purification of the residue by column chromatography (hexane/ethyl acetate ( $3: 1$ )) gave 56 b ( $4.3 \mathrm{~g}, 85 \%$ ): colorless oil; $[\alpha]^{20}{ }_{\mathrm{D}}+4.2$ ( $c$ 2.1, $\mathrm{CHCl}_{3}$ ): IR (film) $\nu_{\text {max }} 3065,3030,2980,2930,2880,2720,1720$, $1595,1450,1375,1370,1350,1300,1245,1215,1170,1160,1115,1065$, $1025,1010,855,745,700 \mathrm{~cm}^{-1}$ : ${ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.15(\mathrm{~d}$, $\left.3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-3-\mathrm{CH}_{3}\right), 1.17\left(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{C}-2^{\prime \prime}-\mathrm{CH}_{3}\right), 1.31$, 1.33, 1.37 and 1.40 (each s. each 3 H , acelonides), $2.02,2.86,3.62,3.80$. 4.02, 4.18-4.34, 4.46 (each $\mathrm{m}, 8 \mathrm{H}, \mathrm{H}-2-6, \mathrm{H}-2^{\prime}, \mathrm{H}-2^{\prime \prime}$ ), AB system ( $\delta_{\mathrm{A}}$ $=4.63, \delta_{\mathrm{B}}=4.71,2 \mathrm{H}, J_{\mathrm{AB}}=11 \mathrm{~Hz}$, benzyl $\left.-\mathrm{CH}_{2}\right), 7.31(\mathrm{mc}, 5 \mathrm{H}$, aryl-H), $\left.9.78(\mathrm{~d}, 1 \mathrm{H}, J=4 \mathrm{~Hz}, \mathrm{CHO}) ;{ }^{13} \mathrm{C} \mathrm{NMR} \mathrm{(63} \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 13.46,15.81,25.27,25.61,26.50,27.90,36.57,50.33,66.41,73.59$, $74.55,76.07,76.50,81.82,108.03,108.96,127.67,127.73,128.33$, 139.10, 203.14; MS (CI, $40^{\circ} \mathrm{C}$ ) m/e 406 (1), 391 (1), 91 (100). Anal. Calcd for $\mathrm{C}_{23} \mathrm{H}_{34} \mathrm{O}_{6}$ : $\mathrm{C}, 67.96 ; \mathrm{H}, 8.43$. Found: $\mathrm{C}, 68.05 ; \mathrm{H}, 8.57$.
( $2 R, 3 S, 4 S, 5 R, 6 R, 7 S$ )-3-O-Benzyl-1,2:6,7-di- $O$-isopropylidene-5-(hydroxymethyl)-4-methyloctane-1,2,3,6,7-pentol (56c). To a stirred suspension of $\mathrm{LiAlH}_{4}(0.30 \mathrm{~g} .7 .90 \mathrm{mmol})$ in diethyl ether ( 50 mL ) was added $56 \mathrm{~b}(5.0 \mathrm{~g}, 12 \mathrm{mmol})$ in ether $(150 \mathrm{~mL})$ dropwise at $0^{\circ} \mathrm{C}$. After stirring for 3 h at $22^{\circ} \mathrm{C}$, salurated aqueous $\mathrm{MgSO}_{4}$ solution ( 1.5 mL ) and powdered $\mathrm{K}_{2} \mathrm{CO}_{3}(70 \mathrm{mg}, 0.5 \mathrm{mmol})$ were added, and the mixture continued to be stirred for 4 h . Filtration over a pad of Celite, followed by drying $\left(\mathrm{MgSO}_{4}\right)$, evaporation of volatiles, and purification of the residue by column chromatography (hexane/ethyl acetate (3:1)) gave $\mathbf{5 6 c}$ ( $4.65 \mathrm{~g}, 95 \%$ ) as a colorless oil: $[\alpha]^{20} \mathrm{D}-14.1\left(c 2.1, \mathrm{CHCl}_{3}\right)$ : IR (film) $\nu_{\max } 3050,2980.2934,2883.1735,1494,1450,1375,1370.1350,1300$,
$1240,1215,1171,1160,1145,1100,1060,1045,1025,1005,990,940$, $860,850,750,735,700 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.98(\mathrm{~d}$, $\left.3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.14(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{H}-8), 1.32,1.34$, 1.39, and 1.43 (each s, each 3 H , acetonides), $1.61(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 2.25$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}-5$ ), 3.20 (dd, $1 \mathrm{H}, J=5 \mathrm{~Hz}, \mathrm{OH}$ ), $3.63-4.31(\mathrm{~m}, 8 \mathrm{H}, \mathrm{H}-1$, $\left.-2,-3, \cdot 5^{\prime},-6,-7\right), \mathrm{AB}$ system ( $\delta_{\mathrm{A}}=4.65, \delta_{\mathrm{B}}=4.78,2 \mathrm{H}, J_{\mathrm{AB}}=11 \mathrm{~Hz}$, benzyl- $\mathrm{CH}_{2}$ ), $7.30\left(\mathrm{mc}, 5 \mathrm{H}\right.$, aryl-H); ${ }^{13} \mathrm{C} \mathrm{NMR}\left(63 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $11.94,15.55,25.12,25.71,26.26,28.21,35.54,39.01,61.15,65.66,73.85$, $75.08,76.32,79.52,81.90,107.35,108.51,127.41,127.48,128.06$, 137.76; MS (EI, $50^{\circ} \mathrm{C}$ ) $\mathrm{m} / \mathrm{e} 408$ (1), 393 (1), $91\left(100,\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}\right)$. Anal. Calcd for $\mathrm{C}_{23} \mathrm{H}_{36} \mathrm{O}_{6}: \mathrm{C}, 67.62 ; \mathrm{H}, 8.88$. Found: $\mathrm{C}, 67.88 ; \mathrm{H}, 8.40$.
( $2 R, 3 S, 4 S, 5 R, 6 R, 7 S$ )-3-O-Benzyl-1,2:6,7-di- $O$-isopropylidene-4-methyl-5-O-((methylsulfonyl)methyl)octane-1,2,3,6,7-pentol (56d). To a stirred solution of $56 \mathrm{c}(2.6 \mathrm{~g}, 6.4 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$ were added $\mathrm{NEt}_{3}(1 \mathrm{~g}, 10 \mathrm{mmol})$ and $\mathrm{CH}_{3} \mathrm{SO}_{2} \mathrm{Cl}(0.8 \mathrm{~g}, 7 \mathrm{mmol})$ dropwise at $0^{\circ} \mathrm{C}$. After stirring for 24 h at $22^{\circ} \mathrm{C}$, brine ( 40 mL ) was added, the organic layer was separated, and the aqueous one was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic phases were dried $\left(\mathrm{MgSO}_{4}\right)$ and concentrated under reduced pressure to furnish after column chromatography (hexane/ethyl acetate ( $1: 1$ )) 56d ( $2.2 \mathrm{~g}, 71 \%$ ) as an unstable colorless oil: ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.10\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C} .4-\mathrm{CH}_{3}\right.$ ), $1.15(\mathrm{~d}, 3 \mathrm{H} . J=7 \mathrm{~Hz}, \mathrm{H}-8), 1.31,1.37,1.41$, and 1.43 (each s , each 3 H , acetonides), 1.56 (m, 1 H, H-4), 2.49 (m, 1 H, H-5), 2.91 (s, 3 H , $\mathrm{S}-\mathrm{CH}_{3}$ ), 3.73, 3.91-4.64 (each m, $8 \mathrm{H}, \mathrm{H}-1,-2,-3,-5^{\prime},-6,-7$ ), AB system $\left(\delta_{\mathrm{A}}=4.63, \delta_{\mathrm{B}}=4.89,2 \mathrm{H}, J_{\mathrm{AB}}=11.5 \mathrm{~Hz}\right.$, benzyl- $\left.\mathrm{CH}_{2}\right), 7.34(\mathrm{mc}, 5$ H, aryl-H).
( $1 S, 2 S, 3 R, 1^{\prime} R, 1^{\prime \prime} R, 2^{\prime \prime} S$ )-1-( $1^{\prime}, 2^{\prime}$ - $O$-Isopropylideneethane- $1^{\prime}, 2^{\prime}$ -diol- $\left.1^{\prime} \cdot y l\right)-3-\left(1^{\prime \prime}, 2^{\prime \prime} \cdot O\right.$ - isopropylidenepropane- $1^{\prime \prime}, 2^{\prime \prime} \cdot$ diol- $\left.1^{\prime \prime}-y l\right)-2$ methyltetrahydrofuran (57). To 56d ( $2 \mathrm{~g}, 4 \mathrm{mmol}$ ) in $\mathrm{MeOH}(20 \mathrm{~mL})$ was added $p-\mathrm{TsOH}$ until the pH became about 3. Concentration of the mixture under reduced pressure and dilution with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$, followed by addition of DMP $(0.62 \mathrm{~g}, 6.0 \mathrm{mmol})$, led to spontaneous cyclization of $56 d$. Saturated aqueous $\mathrm{Na}_{2} \mathrm{SO}_{4}$ solution ( 5 mL ) was added; evaporation of volatiles, dilution with ether $(100 \mathrm{~mL})$, separation of the organic layer, and drying $\left(\mathrm{MgSO}_{4}\right)$ gave after column chromatography (hexane/ethyl acetate (3:1)) $57(0.92 \mathrm{~g}, 75 \%)$; colorless prisms. The material crystallized from $n$-pentane: $\mathrm{mp} 75^{\circ} \mathrm{C} ;[\alpha]^{20} \mathrm{D}-2.5^{(c} 2.1$, $\mathrm{CHCl}_{3}$ ); IR (KBr) $\nu_{\max } 2980,2960,2955,2915,2890,2859,1485,1450$, $1375,1365,1300,1270,1243.1217,1161,1100,1070,1060,1040,1030$, $995,965,925,865,850,810,790,670,640,510 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H}$ NMR (250 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 1.05$ (d, $3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C} \cdot 2-\mathrm{CH}_{3}$ ), $1.24(\mathrm{~d}, 3 \mathrm{H}, J=$ $6.5 \mathrm{~Hz}, \mathrm{H}-3^{\prime \prime}$ ), $1.25,1.33,1.39$ and 1.40 (each s , each 3 H , acetonides), 2.32 (ddq, $1 \mathrm{H}, J=8,7,3.5 \mathrm{~Hz}, \mathrm{H}-2$ ), $2.44(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3), 3.51-4.26$ ( $\mathrm{m}, 8 \mathrm{H}, \mathrm{H}-1,-4,-1^{\prime},-2^{\prime},-1^{\prime \prime},-2^{\prime \prime}$ ) : ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 15.42$, $16.67,25.24,25.67,26.75,28.29,36.53,41.68,67.29,71.09,73.38,77.10$, $77.43,88.05,107.30,109.30$; MS (EI, $\left.80 \mathrm{eV}, 60^{\circ} \mathrm{C}\right) \mathrm{m} / \mathrm{e} 300\left(\mathrm{l},[\mathrm{M}]^{+}\right)$, 285 (5), 199 (31), 141 (63), 97 (39), 57 (100), 43 (92). Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{28} \mathrm{O}_{5}$ : C, $63.97 ; \mathrm{H}, 9.40$. Found: $\mathrm{C}, 64.02, \mathrm{H}, 9.42$.
( $2 R, 3 R, 4 R, 5 R, 6 S, 7 R$ )-3.O-Benzyl-1, 2:6,7-di- $O$-isopropylidene-4-methyl-5-vinyloctane-1,2,3,6,7-pentol (58). To a stirred solution of diol $54 \mathrm{~b}(1.1 \mathrm{~g}, 3.02 \mathrm{mmol})$ and DMP ( $0.49 \mathrm{~g}, 4.7 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ was added $p$ - TsOH until the pH became 3. Workup according to the procedure described above for $\mathbf{5 5}$ gave after purification by column chromatography (hexane/ethyl acetate ( $5: 1$ )) diacetonide 58 ( 1.03 g , 84\%); colorless oil; $[\alpha]^{20} \mathrm{D}+64.2\left(c 1.4, \mathrm{CHCl}_{3}\right)$; IR and MS data see 56a; ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.92$ (d, $3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C} .4-\mathrm{CH}_{3}$ ), $1.16(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{H}-8), 1.34$ and 1.38 (each s, each 3 H , acetonide $-\mathrm{CH}_{3}$ ), $1.45\left(\mathrm{~s}, 6 \mathrm{H}\right.$, acetonide $\left.-\mathrm{CH}_{3}\right), 1.80-1.96(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4)$, 2.79 (ddd, $1 \mathrm{H}, J=8.5,2.5 \mathrm{~Hz}, \mathrm{H}-5$ ), 3.27 (dd, $1 \mathrm{H}, J=9,4 \mathrm{~Hz}, \mathrm{H}-3$ ), 3.83 (dd, $1 \mathrm{H}, J=8,8 \mathrm{~Hz}, \mathrm{H}-1$ ), 4.03 (dd, $\left.1 \mathrm{H}, J=8,7 \mathrm{~Hz}, \mathrm{H}-1^{\prime}\right), 4.16$ (dd, $1 \mathrm{H}, J=8.5,5.5 \mathrm{~Hz}, \mathrm{H}-6), 4.20-4.32(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-2, \mathrm{H}-7), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=4.65, \delta_{\mathrm{B}}=4.74,2 \mathrm{H}, J_{\mathrm{AB}}=12 \mathrm{~Hz}\right.$, benzyl. $\mathrm{CH}_{2}$ ), 5.09 (dd, $1 \mathrm{H}, J=16.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), $5.26(\mathrm{dd}, 1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic $-\mathrm{CH}_{2}$ ), 5.81 (ddd, $1 \mathrm{H}, J=16.5,10,8.5 \mathrm{~Hz}$, vinylic- CH ), 7.34 (mc, 5 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 11.58,16.06,25.40,25.94$, $26.29,28.42,37.81,42.95,66.11,74.09,74.36,77.00,78.78,80.73$, $107.35,108.63,118.30,127.45,128.28,136.06,138.48$; HRMS calcd for $\mathrm{C}_{23} \mathrm{H}_{33} \mathrm{O}_{5}\left(\left[\mathrm{M}-\mathrm{CH}_{3}\right]^{+}\right) 389.2328$, found 389.2329 .

Double Stereodifferentiation; Mismatched Combinations. Compounds 60. (-)-61. 62, and $(+)-61$ are described in ref 22. (2R,3R,4R,5R,6S,7S)-3-O-Benzyl-7-O-((tert-butyldimethyl)silyl)-1,2-O-isopropylidene-4-methyl-5-vinyloctane-1,2,3,6,7-pentol (63a)According to the general procedure, $27(3.37 \mathrm{~g}, 9.13 \mathrm{mmol})$ and $(S)$ - 48 b ( $3.04 \mathrm{~g}, 16.3 \mathrm{mmol}$ ) in THF ( 50 mL each) were treated with a suspension of $\mathrm{CrCl}_{3}(3.91 \mathrm{~g}, 24.7 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.47 \mathrm{~g}, 12.4 \mathrm{mmol})$ in THF ( 100 mL ) at $0^{\circ} \mathrm{C}$ 10 provide after purification by column chromatog. raphy (hexane/erhyl acetate ( $5: 1$ )) $63 \mathrm{a}(2.90 \mathrm{~g}, 66 \%$ ) as a colorless oil; $[\alpha]{ }^{20} \mathrm{D}+28.3$ (c 1.3, $\mathrm{CHCl}_{3}$ ); IR and MS see $53 \mathrm{a} ;{ }^{1} \mathrm{H}$ NMR ( 250 MHz , $\mathrm{CDCl}_{3}$ ) $\delta 0.08$ and 0.11 (each s , each $3 \mathrm{H}, \mathrm{SiCH}_{3}$ ), 0.91 (s, $9 \mathrm{H}, t$ $\left.\mathrm{BuCH}_{3}\right), 1.01\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.06(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}$,

H-8), 1.38 and 1.46 (each s, each 3 H , acetonide), $2.00-2.15$ (m, 1 H , $\mathrm{H}-4), 2.31-2.41(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 2.59(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 3.41(\mathrm{dd}, 1 \mathrm{H}, J=$ $7.5,6 \mathrm{~Hz}, \mathrm{H}-3), 3.42(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6), 3.67(\mathrm{dq}, 1 \mathrm{H}, J=6.5,4 \mathrm{~Hz}, \mathrm{H}-7)$, $3.75(\mathrm{dd}, 1 \mathrm{H}, J=8,8 \mathrm{~Hz}, \mathrm{H}-1), 4.05(\mathrm{dd}, 1 \mathrm{H}, J=8,6 \mathrm{~Hz}, \mathrm{H}-1), 4.35$ (ddd, $1 \mathrm{H}, J=8,8,6 \mathrm{~Hz}, \mathrm{H}-2$ ), AB system $\left(\delta_{\mathrm{A}}=4.64, \delta_{\mathrm{B}}=4.77,2 \mathrm{H}\right.$, $J_{\mathrm{AB}}=12 \mathrm{~Hz}$, benzyl- $\mathrm{CH}_{2}$ ), $4.96\left(\mathrm{dd}, 1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}\right.$, vinylic- $\mathrm{CH}_{2}$ ), 5.15 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.84 (ddd, $1 \mathrm{H}, J=17.5,10$, 10 Hz , vinylic-CH), 7.34 (mc, 5 H , aryl- H ); ${ }^{13} \mathrm{C}$ NMR ( 63 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta-4.80,-4.11,15.01,17.98,19.35,25.60,25.84,26.56,37.24$, $46.43,66.49,70.81,73.41,77.62,82.11,108.81,117.74,12728,127.58$, 128.14, 136.34, 139.02; HRMS calcd for $\mathrm{C}_{27} \mathrm{H}_{46} \mathrm{O}_{5} \mathrm{Si}\left([\mathrm{M}]^{+}\right) 478.31145$, found 478.31143.
( $2 R, 3 R, 4 R, 5 R, 6 S, 7 S$ )-3- $O$-Benzyl-1,2- $O$-isopropylidene-4-methyl5 -vinyloctane-1,2,3,6,7-pentol (63b). According to the desilylation described above for 53 a , silyl ether $63 \mathrm{a}(2.0 \mathrm{~g}, 4.2 \mathrm{mmol})$ was treated with $\mathrm{Bu}_{4} \mathrm{NF}$ (trihydrate) ( $2.06 \mathrm{~g}, 6.5 \mathrm{mmol}$ ) in THF ( 100 mL ) to give after workup and purification by column chromatography (hexane/ethyl acetate ( $1: 1$ )) 63b ( $1.38 \mathrm{~g}, 90 \%$ ) as a colorless oil: IR and MS data see 53b; $[\alpha]^{20} \mathrm{D}+19.8\left(c 1.0, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.02$ $\left(\mathrm{d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.14(\mathrm{~d}, 3 \mathrm{H}, J=6.5 \mathrm{~Hz}, \mathrm{H}-8), 1.37$ and 1.46 (each s, each 3 H , acetonide), 1.89-2.04 (m, $1 \mathrm{H}, \mathrm{H}-4$ ), 2.29-2.42 (m, 1 H, H.5), $2.38(\mathrm{~d}, 1 \mathrm{H}, J=4 \mathrm{~Hz}, \mathrm{OH}), 3.16(\mathrm{~d}, 1 \mathrm{H}, J=4 \mathrm{~Hz}$, $\mathrm{OH}), 3.37(\mathrm{dd}, 1 \mathrm{H}, J=6.5,5.5 \mathrm{~Hz}, \mathrm{H}-3), 3.46(\mathrm{ddd}, 1 \mathrm{H}, J=5.5,4$, $4 \mathrm{~Hz}, \mathrm{H}-6$ ), $3.60-3.76(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-7), 3.68(\mathrm{dd}, 1 \mathrm{H}, J=8,8 \mathrm{~Hz}, \mathrm{H}-1)$, 4.06 (dd, $1 \mathrm{H}, J=8,6.5 \mathrm{~Hz}, \mathrm{H}-1), 4.38$ (ddd, $1 \mathrm{H}, J=8,6.5,6.5 \mathrm{~Hz}$, $\mathrm{H}-2), \mathrm{AB}$ sysiem $\left(\delta_{\mathrm{A}}=4.64, \delta_{\mathrm{B}}=4.88,2 \mathrm{H}, J_{\mathrm{AB}}=12 \mathrm{~Hz}\right.$, benzyl- $\left.\mathrm{CH}_{2}\right)$, $5.05\left(\mathrm{dd}, 1 \mathrm{H}, J=17,2 \mathrm{~Hz}\right.$, vinylic- $\left.\mathrm{CH}_{2}\right), 5.19(\mathrm{dd}, 1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.92 (ddd, $1 \mathrm{H}, J=17,10,8.5 \mathrm{~Hz}$, vinylic- CH ), 7.34 (mc, 5 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 15.73,18.95,25.48,26.57$, $37.68,48.14,66.42,68.56,74.11,75.11,78.24,82.99,109.02,118.37$, $127.61,127.84,128.30,136.67,138.37$; HRMS calcd for $\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{O}_{5}$ ([M $\left.-\mathrm{CH}_{3}\right]^{+}$) 349.20151 , found 349.20156.
( $2 R, 3 R, 4 R, 5 R, 6 S, 7 S$ )-3- $O$-Benzyl-1, 2:6,7-di- $O$-isopropylidene-4-methyl-5-vinyloctane-1,2,3,6,7-pentol (64). To a stirred solution of diol 63b ( $1.1 \mathrm{~g}, 3.02 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{~mL})$ were added DMP $(0.49 \mathrm{~g}$, 4.7 mmol ) and $p-\mathrm{TsOH}$ at $22^{\circ} \mathrm{C}$ until the pH became about 3 . Workup as described for the preparation of $\mathbf{5 5}$ gave after chromatography (hexane/ethyl acetate ( $5: 1$ )) $64(1.0 \mathrm{~g}, 84 \%)$ : colorless oil; IR and MS data see 56a; $[\alpha]^{20}{ }_{\mathrm{D}}+30.7\left(c 30.7, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ $1.02\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.19(\mathrm{~d}, 3 \mathrm{H}, J=6 \mathrm{~Hz}, \mathrm{H}-8), 1.32$, 1.36, 1.38, and 1.46 (each s, each 3 H , acetonides), 2.00 (ddq, $1 \mathrm{H}, J=$ $7.2,7.2,4.2 \mathrm{~Hz}, \mathrm{H}-4$ ), 2.48 (ddd, $1 \mathrm{H}, J=10,4,3 \mathrm{~Hz}, \mathrm{H}-5$ ), 3.34 (dd, $1 \mathrm{H}, J=7,5 \mathrm{~Hz}, \mathrm{H}-3$ ), 3.65 (dd, $1 \mathrm{H}, J=8.5,3 \mathrm{~Hz}, \mathrm{H}-6$ ), 3.76 (dd, $1 \mathrm{H}, J=8,8 \mathrm{~Hz}, \mathrm{H}-1), 3.82(\mathrm{dq}, 1 \mathrm{H}, J=8.5,6 \mathrm{~Hz}, \mathrm{H}-7), 4.05$ (dd, $1 \mathrm{H}, J=8,6.5 \mathrm{~Hz}, \mathrm{H}-1$ ), 4.32 (ddd, $1 \mathrm{H}, J=8,6.5,5 \mathrm{~Hz}, \mathrm{H}-2$ ), AB system $\left(\delta_{\mathrm{A}}=4.63, \delta_{\mathrm{B}}=4.79,2 \mathrm{H}, J_{\mathrm{AB}}=12 \mathrm{~Hz}\right.$, benzyl-CH2), $4.96(\mathrm{dd}$, $1 \mathrm{H}, J=16,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.11 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic$\left.\mathrm{CH}_{2}\right), 5.84$ (ddd, $1 \mathrm{H}, J=16,10,10 \mathrm{~Hz}$, vinylic- CH ), $7.34(\mathrm{mc}, 5 \mathrm{H}$, aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 14.10,17.15,25.40,26.50,26.84$, $27.33,38.81,45.12,66.34,73.50,74.55,77.64,81.85,84.62,107.61$, $108.72,118.62,127.48,128.23,135.47,138.87$; HRMS caled for $\mathrm{C}_{23^{\circ}}$ $\mathrm{H}_{33} \mathrm{O}_{5}\left(\left[\mathrm{M}-\mathrm{CH}_{3}\right]^{+}\right) 389.23280$, found 389.23293 .
(3S,4S,5S,6R,7R )-1,3-O-Benzyl-4-methyl-7-O-( (tert -butyldi-methyl)silyl)-5-vinyloctane-1,3,6,7-tetrol (65). According to the general procedure, $19(1.0 \mathrm{~g}, 2.48 \mathrm{mmol})$ and $(R) .48 \mathrm{~b}(0.70 \mathrm{~g}, 3.74 \mathrm{mmol})$ were treated with a suspension of $\mathrm{CrCl}_{3}(1.06 \mathrm{~g}, 6.69 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(127$ $\mathrm{mg}, 3.35 \mathrm{mmol}$ ) in THF ( 70 mL ) to give after column chromatography (hexane/ethyl acetate ( $5: 1$ )) $65\left(1.15 \mathrm{~g}, 90 \%\right.$ ) as a colorless oil: $[\alpha]^{20}$ $-23.1\left(c 1.8, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.07(\mathrm{~s}, 6 \mathrm{H}$, $\left.\mathrm{SiCH}_{3}\right), 0.89\left(\mathrm{~m}, 12 \mathrm{H}, t-\mathrm{BuCH}_{3}, \mathrm{C} \cdot 4-\mathrm{CH}_{3}\right), 1.00(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}$, $\mathrm{H}-8$ ), 1.79-2.08 (m, 3 H, H-2, H-4), 2.16 (m, $1 \mathrm{H}, \mathrm{H}-5$ ), 2.52 ( $\mathrm{s}, 1 \mathrm{H}$, $\mathrm{OH}), 3.40-3.82(\mathrm{~m}, 5 \mathrm{H}, \mathrm{H}-1,-3,-6,-7), 4.51\left(\mathrm{mc}, 4 \mathrm{H}\right.$, benzyl- $\mathrm{CH}_{2}$ ), 4.93 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), $5.09(\mathrm{dd}, 1 \mathrm{H}, J=10,2$ Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.76 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic-CH), 7.29 (mc, 10 H , aryl-H): ${ }^{13} \mathrm{C} \mathrm{NMR}\left(63 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta-4.77,-4.09,12.84$ 18.03, 19.18. 25.86, 31.73, 36.24, 47.39, 67.60, 71.11, 71.63, 73.03, 75.84 $77.79,117.01,127.31,127.52,127.69,128.24,128.32,137.19,138.59$, 139.15; MS (El, $150^{\circ} \mathrm{C}$ ) m/e 512 (1.26, $[\mathrm{M}]^{+}$), 455 (12.83, [ $\mathrm{M}-$ $t$-Bu] ${ }^{+}$), $131\left(100,[\mathrm{OSi}-t-\mathrm{BuMe}]^{+}\right)$; HRMS; calcd for $\mathrm{C}_{27} \mathrm{H}_{39} \mathrm{O}_{4} \mathrm{Si}([\mathrm{M}$ $-t \cdot \mathrm{Bu}]^{+}$). 455.261762, found 455.261793 .
( $\left.2 S, 3 R / S, 4 R / S, 5 R, 2^{\prime} R / S\right)-1-O$-( tert -Butyldiphenyl) silyI)-2-methyl-5-O-tetrahydropyranyl-3-vinylhexane-1,4,5-triol (66a). According to the general procedure, allylic bromide $13(1.0 \mathrm{~g}, 2.40 \mathrm{mmol})$ and $(R)-48 \mathrm{a}(0.57 \mathrm{~g}, 3.60 \mathrm{mmol})$ were treated with a suspension of $\mathrm{CrCl}_{3}$ $(1.03 \mathrm{~g}, 6.50 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.12 \mathrm{~g}, 3.16 \mathrm{mmol})$ in THF ( 50 mL ) at $0^{\circ} \mathrm{C}$ to give after purification by column chromatography (hexane/ ethyl acetate (3:1)) $66 \mathrm{a}(0.95 \mathrm{~g}, 80 \%)$ as a colorless oil: IR and MS data see 52a; ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.89(\mathrm{~m}, 3 \mathrm{H}, J=7 \mathrm{~Hz}$, $\mathrm{C}-2-\mathrm{CH}_{3}$ ), 1.05 (s, $9 \mathrm{H}, t-\mathrm{BuCH}_{3}$ ), $1.22(\mathrm{mc}, 3 \mathrm{H}, J=6 \mathrm{~Hz}, \mathrm{H}-6)$, 1.42-2.12 (m, 7 H, THP-H-3', $-4^{\prime},-5^{\prime}, \mathrm{H}-2$ ), 2.27 (m, 1 H, H-3), 2.61
(d, $1 \mathrm{H}, \mathrm{OH}$ ), 3.56 (mc, $5 \mathrm{H}, \mathrm{THP}-\mathrm{H}-6^{\prime}, \mathrm{H}-1, \mathrm{H}-4, \mathrm{H}-5$ ), 3.92 (m, 1 H, THP-H-6'), 4.51 and 4.66 (each m, each 1 H, THP-H-2'), 5.04 (mc, 2 H , vinylic- $\mathrm{CH}_{2}$ ), $5.77(\mathrm{~m}, 1 \mathrm{H}$, vinylic- CH$), 7.37(\mathrm{~m}, 6 \mathrm{H}$, aryl- H$), 7.63$ (m, 4 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 14.82,16.25,17.75$, 19.32, 20.17, 20.99, 25.16, 25.40, 26.59, 26.95, 31.20, 36.93, 47.23, 63.20, $66.79,75.60,78.32,100.94,117.66,127.60,129.56,135.61,135.69$, 136.02, 136.31. Anal. Calcd for $\mathrm{C}_{30} \mathrm{H}_{44} \mathrm{O}_{4} \mathrm{Si}: \mathrm{C}, 72.54 ; \mathrm{H}, 8.92$. Found: C, 72.42 ; H, 8.96 .
(2S,3R/S,4R/S,5R)-1-O-((tert-Butyldiphenyl)silyl)-2-methyl-3-vinylhexane-1,4,5-triol ( 66 b ). Deprotection of $66 \mathrm{a}(0.68 \mathrm{~g}, 1.37 \mathrm{mmol}$ ) with $\mathrm{MeOH}(100 \mathrm{~mL})$ and PPTS furnished after column chromatography (hexane/ethyl acetate (2:1)) 66b ( $0.29 \mathrm{~g}, 51 \%$ ) as a colorless oil in a diastereomeric mixture (ratio: 3:1:1): IR and MS data see $\mathbf{5 2 b}$; ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.80$ (d, $J=7 \mathrm{~Hz}$, minor diastereomer $\mathrm{CH}_{3}$ ), $0.84\left(\mathrm{~d}, J=7 \mathrm{~Hz}\right.$, major diastereomer $\left.\mathrm{CH}_{3}\right) 0.85(\mathrm{~d}, J=7 \mathrm{~Hz}$, minor diastereomer $\mathrm{CH}_{3}$ ), $1.05\left(\mathrm{~s}, t-\mathrm{BuCH}_{3}\right), 1.14(\mathrm{~d}, J=6 \mathrm{~Hz}$, major diastereomer $\mathrm{H}-6$ ), 1.24 (d, $J=6 \mathrm{~Hz}$, minor diastereomer $\mathrm{H}-6$ ), 1.25 (d, $J=6 \mathrm{~Hz}$, minor diastereomer H-6), 1.91 (m, H-2), 2.12 (m, major diastereomer $\mathrm{H}-3$ ), 2.39 (m, minor diastereomer $\mathrm{H}-3$ ), 2.66 ( m , minor diastereomer OH ), 3.28 (d, major diastereomer OH ), $3.55(\mathrm{mc}, \mathrm{H}-1,4$, 5), $4.98\left(\mathrm{~m}, J=17.5,2 \mathrm{~Hz}\right.$, vinylic- $\left.\mathrm{CH}_{2}\right), 5.11(\mathrm{~m}, J=10,2 \mathrm{~Hz}$, vinylic- $\left.\mathrm{CH}_{2}\right), 5.80(\mathrm{~m}$, vinylic- CH$), 7.37(\mathrm{~m}$, aryl- H$), 7.63(\mathrm{~m}$, aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.27,15.15,16.48,17.69,18.38,19.14$, 19.29, 26.81, 35.89, 36.84, 37.87, 48.62, 49.63, 66.27, 67.16, 68.48, 68.89, $69.10,74.17,75.76,117.87,118.79,118.89,127.70,129.74,129.77$, $133.03,133.10,133.28,135.24,135.49,135.53,135.61,136.29,136.47$. Anal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{36} \mathrm{O}_{3} \mathrm{Si}$ : $\mathrm{C}, 72.77 ; \mathrm{H}, 8.79$. Found: $\mathrm{C}, 72.33 ; \mathrm{H}$, 8.82.
( $2 R, 3 R, 4 R, 5 R / S, 6 R / S, 7 S, 8 R$ )-1,2:7,8-Di- $O$-isopropylidene-4-methyl-5-vinyldecane-1,2,3,6,7,8-hexol (68). Allylic bromide 27 ( 3.57 g , $9.67 \mathrm{mmol})$ and $67 \mathrm{a}(1.53 \mathrm{~g}, 9.67 \mathrm{mmol})$ were treated with a suspension of $\mathrm{CrCl}_{3}(4.13 \mathrm{~g}, 26.1 \mathrm{mmol})$ and $\mathrm{LiAlH}_{4}(0.50 \mathrm{~g}, 13.2 \mathrm{mmol})$ in THF $(120 \mathrm{~mL})$ to provide after column chromatography (hexane/ethyl acetate ( $3: 1$ )) 68 ( $1.72 \mathrm{~g}, 40 \%$ ) in a diastereomeric mixture of $2.5: 1$. The diastereomers were separated by HPLC ( $0.5 \%$ 2-propanol in hexane). Major diastereomer: colorless oil; $[\alpha]^{20} \mathrm{D}+38.7$ (c0.67); IR (film) $\nu_{\text {max }}$. $3460,3070,3040,2985,2940,2880,1455,1380,1370,1245,1220,1160$, $1060,1000,995,970,950,920,875,860,735,700,515 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.95(\mathrm{t}, 3 \mathrm{H}, J=7.5 \mathrm{~Hz}, \mathrm{H}-10), 1.08(\mathrm{~d}, 3 \mathrm{H}, J$ $=7.5 \mathrm{~Hz}, \mathrm{CH}_{3}$ ), $1.26,1.36,1.43$ (each $\mathrm{s}, 12 \mathrm{H}$, acetonides), $1.50(\mathrm{mc}$, $2 \mathrm{H}, \mathrm{H}-9), 1.89$ (m, $1 \mathrm{H}, \mathrm{H}-4$ ), 2.58 (m, $1 \mathrm{H}, \mathrm{H}-5$ ), 3.40 (dd, $1 \mathrm{H}, \mathrm{J}=$ $7,4.5 \mathrm{~Hz}, \mathrm{H}-3), 3.43(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 3.56(\mathrm{t}, 1 \mathrm{H}, J=8,8 \mathrm{~Hz}, \mathrm{H}-1)$, 3.45-4.08 (m, $4 \mathrm{H}, \mathrm{H}-\mathrm{l}^{\prime}, 6,7,8$ ), 4.39 (ddd, $1 \mathrm{H}, J=8,8,7 \mathrm{~Hz}, \mathrm{H}-2$ ), AB system ( $\delta_{\mathrm{A}}=4.60, \delta_{\mathrm{B}}=4.94,2 \mathrm{H}, J_{\mathrm{AB}}=11 \mathrm{~Hz}$, benzyl- $\mathrm{CH}_{2}$ ), 4.94 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), $5.19(\mathrm{dd}, 1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.99 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic- CH ), 7.33 (mc, 5 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 10.60,17.57,22.71,25.51$, $25.78,26.66,28.38,39.06,48.96,66.55,67.30,74.75,77.50,78.84,79.80$, $84.42,107.34,109.08,118.54,127.65,127.93,128.23,136.84,138.08$; MS (EI, $160 \mathrm{eV}, 130^{\circ} \mathrm{C}$ ), m/e 101 (20.52), $91\left(100,\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}\right), 59$ (32.65), 43 (40.07). Anal. Caled for $\mathrm{C}_{26} \mathrm{H}_{40} \mathrm{O}_{6}: \mathrm{C}, 69.61 ; \mathrm{H}, 8.99$. Found: C, 69.27; H, 9.03.

Minor diastereomer: colorless oil; $[\alpha]^{20} \mathrm{D}+7.2\left(c 0.66, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.96(\mathrm{t}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{H}-10), 1.15(\mathrm{~d}, 3$ $\mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}$ ) $1.34,1.36,1.41,1.44$ (each s , each 3 H , acetonides), 1.50 (mc, $2 \mathrm{H}, \mathrm{H}-9$ ), 2.00-2.25 (m, $2 \mathrm{H}, \mathrm{H}-4, \mathrm{H}-5$ ), 2.30 (d, $1 \mathrm{H}, J=2.5 \mathrm{~Hz}, \mathrm{OH}), 3.42$ (dd, $1 \mathrm{H}, J=6,3 \mathrm{~Hz}, \mathrm{H}-3), 3.67(\mathrm{t}, 1 \mathrm{H}$, $J=8,8 \mathrm{~Hz}, \mathrm{H}-1$ ), $3.92(\mathrm{mc}, 3 \mathrm{H}, \mathrm{H}-6, \mathrm{H} \cdot 7, \mathrm{H}-8), 4.01(\mathrm{dd}, 1 \mathrm{H}, J=$ $8,5 \mathrm{~Hz}, \mathrm{H}-1), 4.30(\mathrm{ddd}, 1 \mathrm{H}, J=8,6,5 \mathrm{~Hz}, \mathrm{H}-2), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=\right.$ $4.60, \delta_{\mathrm{B}}=4.69,2 \mathrm{H}, J_{\mathrm{AB}}=11 \mathrm{~Hz}$, benzyl $\left.-\mathrm{CH}_{2}\right), 5.00(\mathrm{dd}, 1 \mathrm{H}, J=17$, 2 Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.19 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.83 (ddd, $1 \mathrm{H}, J=17,10,10 \mathrm{~Hz}$, vinylic-CH), $7.33\left(\mathrm{mc}, 5 \mathrm{H}\right.$, aryl-H); ${ }^{13} \mathrm{C}$ NMR $\left(63 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 10.70,14.74,22.96,25.86,26.50,28.23,36.18$, $48.04,66.21,68.41,72.69,77.26,77.51,78.35,78.98,81.46,107.67$, $108.95,118.08,127.33,127.68,128.14,137.81,138.87$. Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{40} \mathrm{O}_{6}$ : C, 69.61; H, 8.99. Found: C, 69.69; H, 9.15 .

Synthesis of Nephromopsinic Acid and Its Enantiomer. ( $\mathbf{2 R}, \mathbf{3 S}, \mathbf{4 S}, \mathbf{5 S}, 6 S$ )-3-O-Benzyl-4-methyl-5-vinylnonadecane-1,2,3,6-tetrol (70a). Deprotection of $38(0.53 \mathrm{~g}, 1.05 \mathrm{mmol})$ with $\mathrm{MeOH}(50 \mathrm{~mL})$ and $p-\mathrm{TsOH}$ was accomplished analogously to the procedure described above for 52b to give after purification by column chromatography (hexane/ ethyl acetate (1:1)) 70a ( $0.41 \mathrm{~g}, 84 \%$ ) as a colorless oil: $[\alpha]^{20} \mathrm{D}+1.4$ (c 2.1, $\mathrm{CHCl}_{3}$ ); IR (film) $\nu_{\max } 3400,2930,2860,1480-1190,1150-980$, $920,735,700 \mathrm{~cm}^{-1}:{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.89(\mathrm{t}, 3 \mathrm{H}, J=$ $7 \mathrm{~Hz}, \mathrm{H}-19), 0.99\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.14-1.59(\mathrm{~m}, 24 \mathrm{H}$, H-7-18), 2.08 (m, $1 \mathrm{H}, \mathrm{H}-4$ ), 2.45 (ddd, $1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-5$ ), 2.52 $(\mathrm{m}, 3 \mathrm{H}, \mathrm{OH}), 3.73(\mathrm{mc}, 5 \mathrm{H}, \mathrm{H}-1,2,3,6), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=4.76, \delta_{\mathrm{B}}\right.$ $=4.74,2 \mathrm{H}, J_{\mathrm{AB}}=11.5 \mathrm{~Hz}$, benzyl $\left.\cdot \mathrm{CH}_{2}\right), 5.13(\mathrm{dd}, 1 \mathrm{H}, J=17.5,2$ Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.27 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.82 (ddd, $1 \mathrm{H}, J=17.5 .10,10 \mathrm{~Hz}$, vinylic-CH), $7.34(\mathrm{mc}, 5 \mathrm{H}$, aryl- H$) ;{ }^{13} \mathrm{C} \mathrm{NMR}$
( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.96,14.19,22.54-35.17,50.66,63.80,72.35$ $72.38,73.92,83.13 .118 .45,127.44,128.23,136.90,138.42$; MS (EI, 70 $\left.{ }^{\circ} \mathrm{C}\right) \mathrm{m} / \mathrm{e} 463$ ( 0.57 ), 402 (30.87), 401 (100), 383 (42.36), 365 (24.85), 293 (27.06), 91 (100, $\left[\mathrm{C}_{7} \mathrm{H}_{7}\right]^{+}$), 69 (22.85). Anal. Calcd for $\mathrm{C}_{29} \mathrm{H}_{50} \mathrm{O}_{4}$ : C, $75.28 ; \mathrm{H}, 10.89$. Found: C, $75.45 ; \mathrm{H}, 10.95$.
( $2 R, 3 S, 4 S, 5 S, 6 S$ )-4-Methyl-5-vinylnonadecane-1,2,3,6-tetrol (70b). Debenzylaion of 70 a ( $2.1 \mathrm{~g}, 4.54 \mathrm{mmol}$ ) in THF/ammonia ( 100 mL , 1:1 mixture) with sodium chips, according to the procedure described for the preparation of 44, furnished after column chromatography (ethyl acetate) 70 b ( $1.56 \mathrm{~g} .92 \%$ ) as a colorless oil: IR (film) $\nu_{\max } 3350,3070$. $2920,2840,1460,1415,1375,1325,1050,910,885,725 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.90(\mathrm{t}, 3 \mathrm{H}, \mathrm{H}-19), 0.95(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}$, C.4- $\mathrm{CH}_{3}$ ), $1.17-1.61(\mathrm{~m}, 24 \mathrm{H}, \mathrm{H}-7-18), 1.88(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 2.09(\mathrm{~s}$, $1 \mathrm{H}, \mathrm{OH}$ ), 2.34 (ddd, $1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-5), 2.38(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH})$, $3.00(\mathrm{~s}, 2 \mathrm{H}, \mathrm{OH}), 3.73(\mathrm{~m}, 5 \mathrm{H}, \mathrm{H}-1,2,3,6), 5.18(\mathrm{dd}, 1 \mathrm{H}, J=17.5$, 2 Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.27 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.87 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic-CH); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ $13.15-13.74,22.35-36.25,51.69,71.60,75.92,76.24,76.75,118.43$, 136.52; MS (EI, $80^{\circ} \mathrm{C}$ ) m/e 325 (67.28), 323 (100), 311 (45.46), 111 (100), 69 (45.06). 68 (47.99). Anal. Calcd for $\mathrm{C}_{22} \mathrm{H}_{44} \mathrm{O}_{4}: \mathrm{C}, 70.92 ; \mathrm{H}$, 11.90. Found: C, $70.73 ; \mathrm{H}, 11.76$.
(2R /S,3S,4S,5S)-3-Methyl-2-oxy-5-tridecyl-4-vinyltetrahydrofuran (71). To a stirred solution of tetrol 70b ( $0.92 \mathrm{~g}, 2.47 \mathrm{mmol})$ in diethyl ether ( 100 mL ), $\mathrm{H}_{5} \mathrm{IO}_{6}(1.14 \mathrm{~g}, 5.0 \mathrm{mmol})$ was added at $22^{\circ} \mathrm{C}$. After 1 h of stirring, the mixture was concentrated and diluted with methanol $(100 \mathrm{~mL})$, followed by addition of $\mathrm{K}_{2} \mathrm{CO}_{3}$ until the pH rose above 8 . After stirring for an additional 15 min . Ihe solvent was evaporated, and the residue was diluted with ether ( 70 mL ) and water ( 20 mL ). Extraction of the aqueous layer with ether and drying of the combined etheric phases $\left(\mathrm{MgSO}_{4}\right)$, followed by chromatography of the concentrated residue (hexane/ethyl acetate ( $5: 1$ )), furnished 71 ( $0.56 \mathrm{~g}, 73 \%$ ) as a white amorphous solid: IR (KBr) $\nu_{\max } 3420,3090,2930,2860,1465$, 1105, 995, $920,725 \mathrm{~cm}^{-1}:{ }^{1} \mathrm{H} \operatorname{NMR}\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.89(\mathrm{t}, 6 \mathrm{H}$, $J=7 \mathrm{~Hz}$, tridecyl $\cdot \mathrm{CH}_{3}$ ), $1.00\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C} \cdot 3 \cdot \mathrm{CH}_{3}\right), 1.08(\mathrm{~d}, 3$ $\left.\mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-3-\mathrm{CH}_{3}\right) .1 .17-1.60\left(\mathrm{~m}, 48 \mathrm{H}\right.$, tridecyl- $\left.\mathrm{CH}_{2}\right), 2.00(\mathrm{mc}$, $2 \mathrm{H}, \mathrm{H}-3$ ), $2.48(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 2.69(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 2.96(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=$ $4 \mathrm{~Hz}, \mathrm{OH}), 3.41(\mathrm{~d}, 1 \mathrm{H}, J=5 \mathrm{~Hz}, \mathrm{OH}), 4.08(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 4.28(\mathrm{~m}$, $1 \mathrm{H}, \mathrm{H}-5$ ), 5.10 (mc, 4 H , vinylic- $\mathrm{CH}_{2}$ ), 5.36 (mc, $2 \mathrm{H}, \mathrm{H}-2$ ), 5.76 (mc, 2 H , vinylic-CH); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 11.37,14.05,15.34$, 20.87-32.60, 43.10, 45.81, 51.13, 54.35, 80.83, 81.68, 98.85, 104.83, $116.63,117.28,136.60,136.68$; MS (EI, $\left.40^{\circ} \mathrm{C}\right) \mathrm{m} / \mathrm{e} 310\left(0.52,[\mathrm{M}]^{+}\right)$, 109 (44.89), 98 (95.12), 69 (50.25), 42.9 (100). Anal. Calcd for $\mathrm{C}_{20} \mathrm{H}_{38} \mathrm{O}_{2}: \mathrm{C}, 77.36 ; \mathrm{H}, 12.33$. Found: $\mathrm{C}, 77.53 ; \mathrm{H}, 12.65$.
(3S,4S,5S)-3-Methyl-2-0x0-5-tridecyl-4-vinyltetrahydrofuran (72). $71(0.44 \mathrm{~g}, 1.42 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(5 \mathrm{~mL})$ was added to a stirred suspension of PCC (pyridinium chlorochromate) ( $1.54 \mathrm{~g}, 7.14 \mathrm{mmol}, 5$ equiv) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(200 \mathrm{~mL})$ at $0^{\circ} \mathrm{C}$. After stirring for 2 h at $22^{\circ} \mathrm{C}$, ether ( 200 mL ) and $\mathrm{MgSO}_{4}(2 \mathrm{~g}, 17 \mathrm{mmol})$ were added. Filtration, evaporation of volatiles and purification of the residue by column chromatography (hexane/ethyl acetate (5:1)) furnished $72(0.40 \mathrm{~g}, 92 \%$ ) as colorless crystals: $\mathrm{mp} 28^{\circ} \mathrm{C}$; $[\alpha]^{20} \mathrm{D}-66.6\left(c 1.04, \mathrm{CHCl}_{3}\right)$; IR ( KBr ) $\nu_{\text {max }} 2930,2860,1775$ (lactone- $\mathrm{C}=\mathrm{O}$ ) , 1470, 1460, 1380, 1355, 1320, $1175,1140,1000,975,925 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.90$ $\left(\mathrm{t}, 3 \mathrm{H}, J=7 \mathrm{~Hz}\right.$, tridecyl- $\mathrm{CH}_{3}$ ), $1.22\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-3-\mathrm{CH}_{3}\right), 1.24$ ( $\mathrm{s}, 22 \mathrm{H}, \mathrm{C}-2^{\prime}-\mathrm{C}-12^{\prime}$-tridecyl- $\mathrm{CH}_{2}$ ) $1.51\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}-1^{\prime}\right.$-tridecyl- $\mathrm{CH}_{2}$ ), $2.49(\mathrm{dq}, 1 \mathrm{H}, J=11,7 \mathrm{~Hz}, \mathrm{H}-3), 2.81$ (ddd, $1 \mathrm{H}, J=11,9,8 \mathrm{~Hz}, \mathrm{H}-4$ ), $4.48(\mathrm{~m}, 1 \mathrm{H}, J=8 \mathrm{~Hz}, \mathrm{H}-5), 5.22$ (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic$\mathrm{CH}_{2}$ ), 5.25 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.73 (ddd, $1 \mathrm{H}, J=$ $17.5,10,9 \mathrm{~Hz}$, vinylic- CH ); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.48,14.05$, $22.64-31.89,38.46,51.46,80.92,118.84,133.96,178.68$; MS (EI, 40 $\left.{ }^{\circ} \mathrm{C}\right) m / e 308\left(1.15,[\mathrm{M}]^{+}\right), 68(100)$. Anal. Calcd for $\mathrm{C}_{20} \mathrm{H}_{36} \mathrm{O}_{2}$ : C, $77.87 ; \mathrm{H}, 11.76$. Found: $\mathrm{C}, 77.93 ; \mathrm{H}, 11.58$.
(3S,4S,5S)-4-Carboxy-3-methyl-2-oxo-5-tridecyltetrahydrofuran (Nephromopsinic AcId) ((-)-69). To a stirred solution of lactone 72 (155 $\mathrm{mg}, 0.50 \mathrm{mmol})$ in a solvent mixture of $\mathrm{CCl}_{4}(1 \mathrm{~mL}), \mathrm{CH}_{3} \mathrm{CN}(1 \mathrm{~mL})$, and water ( 1.5 mL ) were added $\mathrm{NaIO}_{4}(439 \mathrm{mg}, 2.05 \mathrm{mmol})$ and ruthenium(III) chloride ( $10 \mathrm{mg}, 0.05 \mathrm{mmol}$ ) at $22^{\circ} \mathrm{C}$. After 2 h at ambient temperature, $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ was added, and the aqueous phase was separated and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The combined organic layers
were filtrated over a pad of Celite. After evaporation of volatiles, the residue was diluted with ether ( 2 mL ) and saturated aqueous $\mathrm{NaHCO}_{3}$ solution ( 2 mL ). The etheric phase was separated, and the aqueous one was acidified with hydrochloric acid ( $3 \mathrm{~mL}, 7 \%$ ). The precipitate was separated by filtration, washed with water $\left(1 \mathrm{~mL}, 0^{\circ} \mathrm{C}\right)$, and gave $(-) .69$ ( $100 \mathrm{mg}, 62 \%$ ) as white crystals, which could be purified by dissolving them in ethanol and dilution with water: $\mathrm{mp} 136^{\circ} \mathrm{C}$ (lit. ${ }^{24} \mathrm{mp} 137^{\circ} \mathrm{C}$ ); $[\alpha]^{20}$ D $-84\left(c 0.25, \mathrm{CHCl}_{3}\right)\left(\right.$ lit. $^{24}[\alpha]^{20} \mathrm{D}-85.1$ ); IR (KBr) $\nu_{\max } 3450$ (carboxyl-OH), 3030, 2980, 2950, 2910, 2840, 1740 (lactone-C $=0$ ), $1465,1415,1355,1330,1245(\mathrm{OH}), 1200(\mathrm{C}-\mathrm{O}), 1180(\mathrm{C}-\mathrm{O}), 1010$, $980,695 \mathrm{~cm}^{-1},{ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.90(\mathrm{t}, 3 \mathrm{H}, J=7 \mathrm{~Hz}$, tridecyl- $\mathrm{CH}_{3}$ ), $1.29\left(\mathrm{~m}, 22 \mathrm{H},\left[\mathrm{C}-2^{\prime}-\mathrm{C}-22^{\prime}\right]\right.$-tridecyl $\left.-\mathrm{CH}_{2}\right), 1.38(\mathrm{~d}, 3 \mathrm{H}$, $J=7 \mathrm{~Hz}, \mathrm{CH}_{3}$ ), $1.58\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{C}-1\right.$ '-tridecyl- $\left.\mathrm{CH}_{2}\right), 3.04(\mathrm{dq}, 1 \mathrm{H}, J=$ $10,7 \mathrm{~Hz}, \mathrm{H}-3$ ), 3.22 (dd, $1 \mathrm{H}, J=10,8.5 \mathrm{~Hz}, \mathrm{H}-4$ ), $4.70(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5$ ), $5.50\left(\mathrm{~s}\right.$, br 1 H , carboxyl-OH); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 14.02$, 14.39, 22.62-31.86, 36.44, 51.56, 77.31, 175.00, 177.34 (carboxyl-C); MS (EI, $120^{\circ} \mathrm{C}$ ) m/e 326 (6.01, [M] ${ }^{+}$) 117.0 (40.19), 99 (26.34), 56 (32.72), 55 (83.82). 44 (53.6), 43 (100), 40.8 (80.69). Anal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{34} \mathrm{O}_{4}: \mathrm{C}, 69.90 ; \mathrm{H}, 10.50$. Found: $\mathrm{C}, 69.97 ; \mathrm{H}, 10.71$.
( $\mathbf{2 R}, 3 \boldsymbol{R}, 4 \boldsymbol{R}, 5 R, 6 R$ )-3- $O$-Benzyl-4-methyl-5-vinylnonadecane-1, 2,3,6tetrol (73a). Deprotection of $40(0.53 \mathrm{~g}, 1.05 \mathrm{mmol})$ with $\mathrm{MeOH}(50$ mL ) and $p-\mathrm{TsOH}$ as described furnished after column chromatography (hexane/ethyl acetate (1:1)) 73a ( $0.4 \mathrm{~g}, 85 \%$ ) as a white amorphous solid: IR and MS data see 70a; ${ }^{1} \mathrm{H}$ NMR ( $250 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 0.87(\mathrm{t}, 3 \mathrm{H}$, $J=7 \mathrm{~Hz}, \mathrm{H}-19), 0.97\left(\mathrm{~d}, 3 \mathrm{H}, J=7 \mathrm{~Hz}, \mathrm{C}-4-\mathrm{CH}_{3}\right), 1.16-1.60(\mathrm{~m}, 24$ H, H-7-18), 1.68 (d, $1 \mathrm{H}, J=4 \mathrm{~Hz}, \mathrm{OH}$ ), $2.20(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4), 2.28$ (s, $1 \mathrm{H}, \mathrm{OH}$ ), 2.28 (ddd, $1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-5$ ), $2.54(\mathrm{~d}, 1 \mathrm{H}, J=8$ $\mathrm{Hz}, \mathrm{OH}), 3.40(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3), 3.61$ (mc, $3 \mathrm{H}, \mathrm{H}-1,6), 3.78(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{H}-2), \mathrm{AB}$ system $\left(\delta_{\mathrm{A}}=4.53, \delta_{\mathrm{B}}=4.75,2 \mathrm{H}, J_{\mathrm{AB}}=11 \mathrm{~Hz}\right.$, benzyl $\left.-\mathrm{CH}_{2}\right)$, 5.14 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.29 (dd, $1 \mathrm{H}, J=10,2$ Hz , vinylic- $\mathrm{CH}_{2}$ ), 5.80 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic-CH), 7.33 (mc, 5 H , aryl-H); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 12.63,14.04,22.62$, $25.47,29.29,29.62,31.86,34.82,35.45,50.84,65.03,71.47,73.64,80.71$, $119.53,127.61,127.83,128.44,136.33,137.94$.
( $2 R, 3 R, 4 R, 5 R, 6 R$ )-4-Methyl-5-vinylnonadecane-1,2,3,6-tetrol (73b). According to the debenzylation of 70a, benzyl ether 73a ( $1.30 \mathrm{~g}, 2.81$ mmol) in THF/ammonia ( $100 \mathrm{~mL}, 1: 1$ mixture) was deprotected by addition of sodium. Usual workup gave after column chromatography (ethyl acetate) 73b ( $0.98 \mathrm{~g}, 94 \%$ ) as a white amorphous solid: IR and MS data see 70b; ${ }^{1} \mathrm{H}$ NMR $\left(250 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.88\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{C}-4-\mathrm{CH}_{3}\right.$ and H-19), 1.12-1.60 (m, $24 \mathrm{H}, \mathrm{H}-7-18$ ), 1.99 (m, 1 H, H-4), 2.29 (ddd, $1 \mathrm{H}, J=10,5,5 \mathrm{~Hz}, \mathrm{H}-5), 3.11(\mathrm{~d}, 1 \mathrm{H}, \mathrm{OH}), 3.40(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-6), 3.71$ (mc, $4 \mathrm{H}, \mathrm{H}-1,-2,-3), 4.04(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 4.10(\mathrm{~d}, 1 \mathrm{H}, J=5 \mathrm{~Hz}, \mathrm{OH})$, 4.22 (s. $1 \mathrm{H}, \mathrm{OH}$ ), 5.12 (dd, $1 \mathrm{H}, J=17.5,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.21 (dd, $1 \mathrm{H}, J=10,2 \mathrm{~Hz}$, vinylic- $\mathrm{CH}_{2}$ ), 5.78 (ddd, $1 \mathrm{H}, J=17.5,10,10 \mathrm{~Hz}$, vinylic-CH); ${ }^{13} \mathrm{C}$ NMR ( $63 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 13.38,14.03,22.63,25.93$, $29.31,29.68,31.88,34.72,36.75,51.75,65.81,70.54,71.94,75.25$, $118.96,136.55$. Anal. Calcd for $\mathrm{C}_{22} \mathrm{H}_{44} \mathrm{O}_{4}: \mathrm{C}, 70.92 ; \mathrm{H}, 11.90$. Found: C, 71.31; 11.87. The synthesis of ent-71, ent-72, and ent-69 was accomplished in accordance with the preparation described above for 71,72, and (-)-69. The chromatographic (HPLC) and spectroscopic ( ${ }^{1} \mathrm{H} /{ }^{13} \mathrm{C}$ NMR, IR, MS) properties of these compounds were identical, respectively, and differed only in the sign of the optical rotation.

Acknowledgment. This work was generously supported by the Schering AG, Berlin-Berkamen, the Fonds der Chemischen Industrie, and the Graduiertenkolleg "Synthese und Strukturaufklärung niedermolekularer Verbindungen". We also thank Dr. B. Kirste and W. Münch for the NOE measurements.

Supplementary Material Available: Procedures and analytical data ( ${ }^{1} \mathrm{H} /{ }^{13} \mathrm{C}$ NMR, IR, MS spectra, optical rotations, combustion analysis) of compounds not described in the Experimental Section, NOE experiments for $\mathbf{4 3}, 55,58$, and 64 , and tables of atomic coordinates and bond angles and distances for 44 and 57 (48 pages); tables of observed and calculated crystal structure factors for 44 and 57 ( 26 pages). Ordering information is given on any current masthead page.


[^0]:    ${ }^{\dagger}$ Inslitul für Organische Chemie der Freien Universilảı Berlin.
    ! Insliluı für Krisıallographie der Freien Universitả Berlin.
    $\ddagger$ Preparalive work.
    ${ }^{\perp}$ Crysial siruclure analysis of compounds 44 and 57.

[^1]:    (12) (a) Chèrest, M.; Felkin, H.; Prudent, N. Tetrahedron Lett. 1968 2199. Anh, N. T. Top. Curr. Chem. 1980, 88, 145. Cram, D. J.: Abd Elhafez, F. A. J. Am. Chem. Soc. 1952, 74, 5828. (b) Houk, K. N.; Paddon-Row, M. N.; Rondan, N. G.; Wu, Y.-D.; Brown, F. K.; Spellmeyer, D. C.; Melz, J. T.; Li. Y.; Loncharich, R. J. Science 1986, 231, 1108 . Wu. Y.-D.; Houk, K. N. J. Am. Chem. Soc. 1987, 109, 908. Li, Y.; Paddon-Row, M. N.; Houk, K N. J. Am. Chem. Soc. 1988, 110,3684 . (c) Hoffmann, R. W. Chem. Rev. 1989, 89, 1841. Reviews: Mulzer, J. Nachr. Chem. Tech. Lab. 1984, 32, 16. Reelz, M. T. Angew. Chem. Int. Ed. Engl. 1984, 23, 556. Eliel, E. L. In Asymmetric Synthesis, Morrison, J. D., Ed.; Academic Press, New York 1983; Vol. 2, p 125.
    (13) Roush, W. R.; Palkowilz, A. D.; Palmer, M. A. J. J. Org. Chem. 1987, 52, 316
    (14) Nagaoka, H.; Kishi, Y. Tetrahedron 1981, 37, 3873.
    (15) Mulzer, J.; Autenrieth-Ansorge, L.; Kirstein, H.: Matsuoka, T.; Münch, W. J. Org. Chem. 1987, 52, 3784.
    (16) Mulzer, J.; Salimi, N. Liebigs Ann. Chem. 1986, 1172.

[^2]:    (17) Winterfeldt, E. Prinzipien und Methoden der Stereoselektiven Synthese; F. Vieweg \& Sohn: Verlagsgesellschaft/Braunschweig, 1988; P 1 .
    (18) Zimmerman, H. E.; Traxler, M. D. J. Am. Chem. Soc. 1957, 79, 1920.
    192.

[^3]:    (20) Mulzer. J.: Schulze. T.; Sirecker, A.: Denzer, W. J. Org. Chem. 1988, 53.4098.
    (21) Horeau, A.; Kagan, H. B.: Vigneron, J. P. Bull. Soc. Chim. Fr. 1968, 3795. Review: Masamune. S.; Choy, W.; Petersen, J. S.; Sita. L. R. Angew. Chem., Int. Ed. Engl. 1985, 24, 1.

[^4]:    (22) Mulzer, J.; Kaltner, L. Angew. Chem., Int. Ed. Engl. 1990, 29, 679.
    (23) Fronza, G.; Fuganti, C.: Grasselli, P.; Pedrocchi-Fantoni, G.; Zirolti, C. Chem. Lett. 1984, 335.

[^5]:    (26) Massad, S. K.; Hawkins, L. D.; Baker, D. C. J. Org. Chem. 1983, 48, 5180.

